

STOICHIOMETRY

This section explains how to calculate the number of moles required or produced, based on a stoichiometric principle. It teaches you how to interpret chemical equations for units of measuring moles, representative particles, masses, and volumes at STP.

1. How can you determine the quantities of reactants and products in a chemical reaction?

*All prices include delivery and handling charges. Prices are subject to change without notice.

2. Quantity usually denotes the _____ amount of a substance expressed in grams or moles.
3. A hexo-ketone is in the family from 3 aldehydes (58%), 2 cyclic ketones (58%), 1 top-TC, 1 ketone (54%), and 4 alcohols (1). Printer is "boldfaced equation" for the construction of this reference.

2009 = 2008 + 1 = 2009

4. In the following sentence, insert a subject, a verb, and a complement in the indicated positions.

- 46.** Chlorophyll levels during flowering in *Silene alba* are not correlated with growth rate. [doi:10.1371/journal.pone.0089409.g004](#)

doi:10.1371/journal.pone.0142011.g001


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7. Circle the letter of the term that tells what kind of information your CLAIM gets from a chemical equation.

- a. radius
b. mass
c. size of particles
d. volume
e. number of particles

Section 12 The Arithmetic Of Equations Practice Problems

Samuel H. WINTER



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