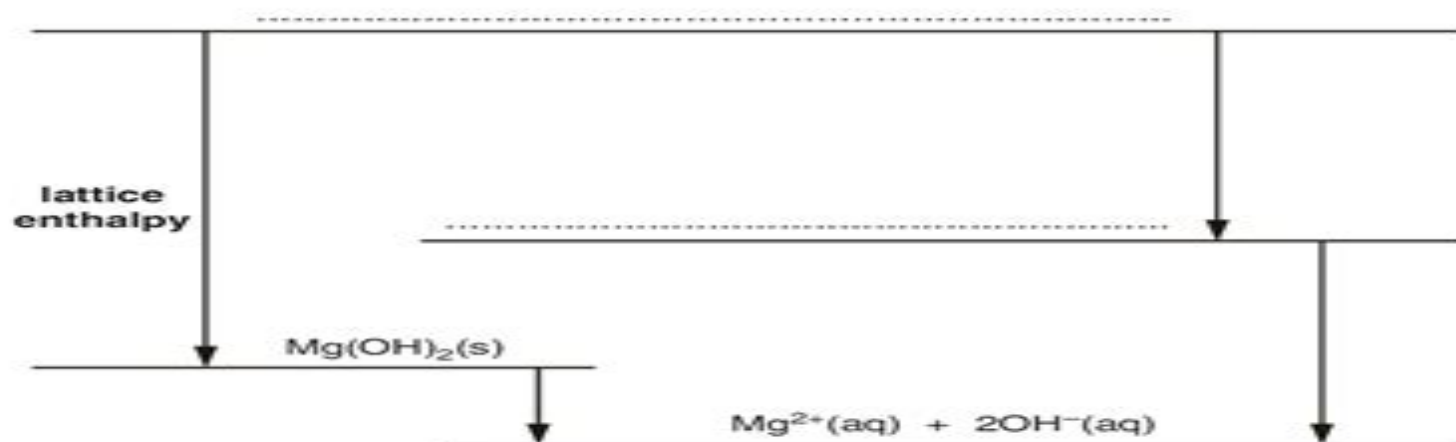


- (iii) A Born–Haber cycle can be drawn to link the lattice enthalpy and enthalpy change of solution of  $\text{Mg}(\text{OH})_2$  with hydration enthalpies.

On the two dotted lines, add the species present, including state symbols.



[2]

- (iv) Calculate the lattice enthalpy of  $\text{Mg}(\text{OH})_2$ .

lattice enthalpy = .....  $\text{kJ mol}^{-1}$  [2]

# Ocr 17 Jan 2013 Canswer Paper

**Vesna Kadelburg, Ben Woolley**



## Ocr 17 Jan 2013 Canswer Paper:

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