#### California State Polytechnic University, Pomona Dr. Laurie S. Starkey, CHM 317L/318L/319L/422L Completing the Reagent Table for Prelab

A Reagent Table is an important part of a lab notebook. It contains the amounts of each "ingredient" used in an experiment (starting material, reagents, catalysts, solvents, etc.), as well as relevant physical properties for each species. The Amounts of each will be given in the experimental procedure, but you must use reference materials such as the CRC Handbook or Wikipedia to find the physical properties of each reagent used in the experiment. To complete the Table, you must calculate the moles, determine the limiting reagent and calculate the theoretical yield (TY) of product - the maximum amount that can be formed. The molar equivalents column provides a simple ratio of the various reagents (limiting reactant is set to 1.0 equiv.) This part of the prelab is to be completed prior to the BEGINNING of each lab. The sample table below is provided for practice. You need to know that each component reacts stoichiometrically (1:1) in the given reaction.

Br.	NH.	1) NaNO <sub>2</sub> , HCI 2) KI		→ Br			7 Welnde here a	
Reagent	/mol	9/mL Density	Amount	Mole	Equiv.	Remarks	a is lud	morely as
2-bromoaniline	172	1.56	2.0 mL	0.018	1.0	bp 229 °C	/ Incan	معروب مع معروب و مع معروب و معروب
SNaNO2	69		1.9 g	0.028	1.6	-	> am	of broken
HCI (aq.)			5.0 mL		excess	Corrosive!	( ).mak	مؤدة وكما
CKI	166		3.6 g	0.022	1.2	-	1 600	a silah
→ H <sub>2</sub> O			10 mL	-		-	1 (mb)	Ser
2-iodobromo- benzene Show your work	283 here:	2.2	Tage.	APRIL .	han fre	Care S		
► benzene	here:	1.56g 1 mc 1 mol 699	* 17	mod - 129	noting r	mp9°C	2-brev	nomid

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Anju Rawlley, D. V. Saraf

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**Fundamentals of General, Organic, and Biological Chemistry** John R. Holum, Ruth C. Denison, 1978 **Microscale Inorganic Chemistry** Zvi Szafran, Ronald M. Pike, Mono M. Singh, 1991-01-24 A comprehensive treatment of the subject of microscale inorganic chemistry is provided through 45 laboratory experiments These include experiments in main group and transition metal chemistry instrumental techniques kinetics synthesis and the manipulation of air sensitive material

Laboratory Manual for Fundamentals of General, Organic, and Biological Chemistry, Third Edition John R. Holum, Ruth C. Denison, Sandra L. Olmsted, 1986 This Laboratory Manual is designed to accompany the texts Fundamentals of General Organic and Biological Chemistry 2nd Edition and Elements of General and Biological Chemistry 6th Edition by John R Holum It is also appropriate for any one year course treating a survey of chemistry at this level and for one term courses covering the whole spectrum of any part of it The experiments have been used by students and have been frequently revised following student polls regarding clarity and interest and suggestions from instructors The questions on the Report and Observation Sheets have again been adjusted in the light of student comments and more room for answers has been provided **Chemistry** McGraw-Hill Staff,2001-07 Experimental Organic Chemistry Joaquín on many Report Sheets Isac-García, José A. Dobado, Francisco G. Calvo-Flores, Henar Martínez-García, 2015-10-30 Experimental Organic Chemistry Laboratory Manual is designed as a primer to initiate students in Organic Chemistry laboratory work Organic Chemistry is an eminently experimental science that is based on a well established theoretical framework where the basic aspects are well established but at the same time are under constant development Therefore it is essential for future professionals to develop a strong background in the laboratory as soon as possible forming good habits from the outset and developing the necessary skills to address the challenges of the experimental work This book is divided into three parts In the first safety issues in laboratories are addressed offering tips for keeping laboratory notebooks In the second the material the main basic laboratory procedures preparation of samples for different spectroscopic techniques Microscale Green Chemistry and qualitative organic analysis are described The third part consists of a collection of 84 experiments divided into 5 modules and arranged according to complexity The last two chapters are devoted to the practices at Microscale Synthesis and Green Chemistry seeking alternatives to traditional Organic Chemistry Organizes lab course coverage in a logical and useful way Features a valuable chapter on Green Chemistry Experiments Includes 84 experiments arranged according to increasing complexity Applied Chemistry | AICTE Prescribed Textbook - English Anju Rawlley, D. V. Saraf, 2021-11-01 This text book o Applied Chemistry is development as per AICTE model curriculum 2018 for compulsory course on Applied Chemistry of first years Diploma Programme in Engineering and Technology Atomic Structure Chemical Bonding Solution Water Engineering Materials Chemistry of fuels Lubricants and Electrochemistry are the five units of this book comprising of both practicals and theory Some salient features of the book l Course Outcomes and Unit Outcomes are written specifically and are mapped with programme Outcomes I Utmost care have been taken to amalgamate the philosophy of outcome based

education l The structure of the textbook is comprehensive where in practical exercises are integral part of each unit l The text is presented in a very simple way with illustrations examples tables flow chart self assessment questions and their solutions l Micro projects points issue for the creative inquisitiveness curiosity know more video links case study and summary points are integral part of each unit to facilitate the students to develop the attitude of scientific inquiry investigate the cause and effect relationship systematic scientific logical thinking ability to observe analyse and interpret l To meet the requirement of outcome based education OBE and outcome based assessment OBA criterion referenced testing CRT have been used as an integral part of assessment in each practical l Sample QR codes have been provided in each units on some topics sub topics for supplementary reading and reinforcing the learning **Green Organic Chemistry** Kenneth M. Doxsee, James E. Hutchison, 2004 This lab text describes the tools and strategies of green chemistry and the lab experiments that allow investigation of organic chemistry concepts and techniques in a greener laboratory setting Students acquire the tools to assess the health and environmental impacts of chemical processes and the strategies to improve develop new processes that are less harmful to human health and the environment The curriculum introduces a number of state of the art experiments and reduces reliance on expensive environmental controls such as fume hoods Provided by publisher

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