

Organic chemistry II (alkanoic acids and alkanols)

- Ethylbutanoate
 - $\text{CH}_3\text{CH}_2\text{CH}_2$
 - Esters
- $-\text{CH}-\text{CH}-\text{CH}_2-\text{CH}-\text{CH}_2-\text{CH}-\text{CH}_2-\text{CH}$
 - Polyphenyl ethane
- Plastics may contain chlorine or fluorine compounds apart from hydrogen and carbon when burnt, fluorine and chlorine compounds are released into the air destroying Ozone layer
- $(\text{NH}_4)_2\text{CO}_{(aq)} \quad 2\text{NH}_{3(g)} + \text{CO}_{2(g)} + \text{H}_2\text{O}_{(l)}$
- The first amount of soap precipitates $\text{Ca}^{2+}_{(aq)}$ and $\text{Mg}^{2+}_{(aq)}$ ions and soften water. Then additional soap dissolves oil from the fabric.
- $$\begin{array}{ccccccc} \text{CH}_3\text{CH}_2 & \text{O} & & \text{O} & \text{CH}_2\text{CH}_3 & \text{O} & \\ | & & & & | & & \\ -\text{NH}-\text{CH} & - & \text{C} & - & \text{NH}-\text{C} & - & \text{C} & - & \text{C}- \end{array}$$

$$\begin{array}{ccc} \text{H} & & \text{H} \\ | & & | \\ \text{b) } 0.00005\text{mol. P} = 0.515 \text{ g of monomer.} \\ - 1.0 \text{ mole of poly mer} = \frac{1 \times 0.515}{0.0005} = 10300 \text{ g} \end{array}$$

$$\begin{aligned} \text{RFM (C}_4\text{H}_9\text{ND}_2)_n &= 48 + 9 + 32 = 103 \\ &= (\text{C}_4\text{H}_9\text{NO}_2) = 10300 \\ 103n &= 10300 \\ \therefore n &= 100 \text{ molecules} \end{aligned}$$

- Agent A – magnesium salt formed is soluble hence doesn't form scum
- (a) Styrene/Phenylethene

(b) Addition polymerization

(c) – can be made into different shapes easily

- are cheaper
 - are not corroded by acids, alkalis or air
 - are stronger and long lasting
 - are water-proof
- Any 1 correct

- Add water to the mixture and shake where ethanol dissolves in water while pentane is immiscible.

*MAT

- Transfer the mixture in a separating funnel and allow it to settle when pentane floats on top of water-ethanol mixture.

*MAT

- Turn on the tap to collect water-ethanol mixture while pentane remains in the separating funnel.
- Separate ethanol from water by fractional distillation based on the differences in boiling points.

- Is 100% ethanol/is pure ethanol without water in it
 - 30°C and yeast

✓ 1

- $R = \frac{\Delta x}{\Delta t}$
- $= \frac{43 - 40.5}{}$

Organic Chemistry Que And Ans

Jin-Ying Zhang



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