

19

ACIDS, BASES, AND SALTS

SECTION 19.1 ACID-BASE THEORIES (Pages 587-593)

This section compares and contrasts acids and bases as defined by the theories of Arrhenius, Brønsted-Lowry, and Lewis. It also introduces conjugate acid-base pairs in acid-base reactions.

► Properties of Acids and Bases (pages 587-588)

1. Circle the letters of all the terms that complete the sentence correctly.

The properties of acids include _____.

- a. reacting with metals to produce oxygen
- b. giving foods a sour taste
- c. forming solutions that conduct electricity
- d. causing indicators to change color

2. Bases are compounds that react with acids to form _____ and a(n) _____.

3. Circle the letters of all the terms that complete the sentence correctly.

The properties of bases include _____.

- a. tasting bitter
- b. feeling slippery
- c. changing the color of an indicator
- d. always acting as a strong electrolyte

4. What are the three main definitions of acids and bases? _____

5. Complete the table about the three definitions of acids and bases. (pg. 592)

Acid-Base Definitions		
Type	Acid	Base
Arrhenius-Lowry		H ⁺ acceptor
	electron-pair acceptor	
	H ⁺ producer	

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