

1. In the list below, circle the properties of acids and box the properties of bases.

Sour taste

Feel slippery

Contains hydroxide ions

Turns litmus blue

Reacts with metals to form H<sub>2</sub>

Turns litmus red

Bitter taste

Contains hydrogen ions

Corrosive

2. Complete the chart below:

### Acid-Base Definitions

Type	Acid Definition	Base Definition
Arrhenius	Donates an H <sup>+</sup> ion in solution	Donates an OH <sup>-</sup> ion in solution
Bronsted-Lowry	Donates an H <sup>+</sup> ion	Accepts an H <sup>+</sup> ion
Lewis	Accepts an electron pair	Donates an electron pair

3. What term(s) describe a substance that can act as an acid or a base? **Amphoteric or amphiprotic**
4. Neutralization reactions always produce a **SALT** and **WATER**

5. If the [H<sup>+</sup>] is greater than the [OH<sup>-</sup>], the solution is

acidic

basic

neutral

6. If the [OH<sup>-</sup>] is greater than the [H<sup>+</sup>], the solution is

acidic

basic

neutral

7. If the [H<sup>+</sup>] is equal to the [OH<sup>-</sup>], the solution is

acidic

basic

neutral

8. If the pH is 7, the solution is

acidic

basic

neutral

9. If the pH is greater than 7, the solution is

acidic

basic

neutral

10. If the pH is less than 7, the solution is

acidic

basic

neutral

11. How many moles of lithium hydroxide are required to neutralize 19.2 grams of chloric acid?



$$(19.2 \text{ g HClO}_3) \times (1 \text{ mol HClO}_3 / 84 \text{ g HClO}_3) \times (1 \text{ mol LiOH} / 1 \text{ mol HClO}_3) = 0.228 \text{ mol LiOH}$$

12. How many grams of calcium hydroxide are required to neutralize 132.4 grams of nitric acid?



$$(132.4 \text{ g HNO}_3) \times (1 \text{ mol HNO}_3 / 63 \text{ g HNO}_3) \times (1 \text{ mol Ca(OH)}_2 / 2 \text{ mol HNO}_3) \times (74 \text{ g Ca(OH)}_2 / 1 \text{ mol Ca(OH)}_2) \\ = 77.76 \text{ g Ca(OH)}_2$$

## Section 20 Acids And Bases Answers

**Donald A. McQuarrie, Stanley Gill**

**Section 20 Acids And Bases Answers:**

## **Section 20 Acids And Bases Answers** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Section 20 Acids And Bases Answers**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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