

## II PUC MID-TERM EXAMINATION, OCTOBER - 2023

Time : 3 Hrs. 15 Mins.

SUBJECT : CHEMISTRY (30)

Max Marks : 70

### Instructions

- 1) The question paper has FIVE parts. All parts are compulsory.
- 2) a) Part-A carries 10 marks. Each question carries 1 mark.  
b) Part-B carries 16 marks. Each question carries 2 marks.  
c) Part-C carries 15 marks. Each question carries 3 marks.  
d) Part-D carries 20 marks. Each question carries 5 marks.  
e) Part-E carries 09 marks. Each question carries 3 marks.
- 3) In Part-A, questions first attempted answer will be considered for awarding marks.
- 4) Write balanced chemical equations and draw labeled diagrams, graphs wherever required.
- 5) Direct answers to the numerical problems without detailed steps and specific units for final answer will not carry any marks.
- 6) Use log tables and simple calculations if necessary. (Use of Scientific Calculator is not allowed)

### PART - A

1x15=15

- 1) Select the correct option from the given choices.
- 1) Solubility of a substance is depends on  
a) Nature of solute                      b) Temperature                      c) pressure                      d) All of these
- 2) Raw Mangoes shrivel into pickles when placed in the concentrated solution of NaCl due to  
a) Osmosis                      b) Reverse osmosis  
c) Osmotic pressure                      d) Lowering of vapour pressure
- 3) The charge required to liberate one gram equivalent of an element is  
a) 96,500 F                      b) 1 F                      c) 1 C                      d) 10 F
- 4) Chemical composition of 'Rust' is  
a)  $\text{FeO} \cdot 2\text{H}_2\text{O}$                       b)  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$                       c)  $\text{Fe}_3\text{O}_4$                       d)  $\text{FeO}$
- 5) Unit of molar conductivity is  
a)  $\text{sm}^2 \text{mol}^{-1}$                       b)  $\text{scm}^{-1} \text{mol}^{-1}$                       c)  $\text{sm} \text{mol}^{-1}$                       d)  $\text{mol}^{-1} \text{m}^2$
- 6) In 3d series element which shows largest number of oxidation state is  
a) Cr                      b) Mn                      c) Fe                      d) V
- 7) With respect to actinoids, which of the following statement is not correct  
a) All are radioactive                      b) Diamagnetic in nature  
c) Highly active in the finely divided state                      d) Silvery metals
- 8) The ligand  $\text{SCN}^-$  is an example for  
a) Monodentate                      b) Didentate                      c) Tridentate                      d) Ambidentate
- 9) An example for heteroleptic complex is  
a)  $[\text{Co}(\text{NH}_3)_4\text{Cl}_2]$                       b)  $[\text{CoCl}_2(\text{en})_2]$                       c)  $\text{Pt}(\text{NH}_3)_4\text{Cl}_2$                       d) All the above
- 10) In the reaction  $\text{R}-\text{Cl} + \text{KCN} \rightarrow \text{R}-\text{CN} + \text{KCl}$ . To obtain ethanenitrile  $\text{R}-\text{Cl}$  must be  
a) Chloro Ethane                      b) 1-Chloropropane                      c) Chloro methane                      d) 2-Chloro propane
- 11) The following reaction is known as  $\text{CH}_3\text{Br} + \text{NaI} \xrightarrow{\text{acetone}} \text{CH}_3\text{I} + \text{NaBr}$   
a) Sandmeyer's reaction                      b) Swart's reaction  
c) Friedel Craft's reaction                      d) Finkelstein reaction
- 12) The alcohol in which  $-\text{OH}$  group attached to  $\text{SP}^2$  hybridised carbon atom present next to the carbon-carbon double bond is  
a) Benzylic alcohol                      b) Vinylic alcohol                      c) Allylic alcohol                      d) None of the above
- 13) The IUPAC name of 'wood spirit' is  
a) Methanol                      b) Ethanol                      c) Methanal                      d) Ethanal

(11T02)

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