



Seat No.	
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**T.Y. B.Sc. (Semester – III) Examination, 2013**  
**CHEMISTRY (Paper – IV)**  
**CH-334 : Analytical Chemistry**  
**(2008 Pattern) (New Course)**

Time : 2 Hours

Max. Marks : 40

- N.B.:-** 1) All questions are compulsory.  
2) Figures to the right indicate full marks.  
3) Use of log tables and calculators is allowed.  
4) Neat diagram must be drawn wherever necessary.

1. Answer the following :

10

- 1) Define the term solubility product.
- 2) State Beer's law.
- 3) Define the term  $\lambda_{max}$ .
- 4) What is meant by chemical interference in AAS?
- 5) What is principle of AAS?
- 6) Give the expression to calculate the number of excited molecules in FES.
- 7) What is spectral interference in FES?
- 8) Calculate the transmittance of the solution having turbidance 0.480.
- 9) State Faraday's second law of electrolysis.
- 10) Define the term electrochemical equivalent.

2. a) Answer any two of the following :

6

- 1) Write note on electrolytic separation of copper and nickel.
- 2) Explain spectrophotometric titrations.
- 3) What is criterion for deciding whether turbidimetry or nephelometry should be used in analysis of turbid solution?

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