

PCMH- The Mole Concept (How can atoms be counted?)

Play!

Consider the following equation for a chemical reaction of 16 cells + 40 cells → 40 cells
 Chlorine is represented by one molecule of hydrogen and one molecule of oxygen (molecules)
 from the above molecules. But how can we determine the number of molecules of each molecule?
 Usually, a reaction involves with the stoichiometric number of molecules. How many molecules of each
 substance are there? The reaction can be written as follows: $2H_2 + O_2 \rightarrow 2H_2O$
 The answer is the mole (the number of molecules) of each substance.

Mole-1: The Balanced Equation

(How can the molecules be represented by a balanced equation for the reaction?)

Consider the reaction:



Key Questions

1. What is the balanced equation for the following chemical reaction?

Final Task

Consider the following chemical reaction of 16 cells + 40 cells → 40 cells
 from the above molecules. But how can we determine the number of molecules of each molecule?

1. Consider the following chemical reaction of 16 cells + 40 cells → 40 cells.

PARTICULAR

Reaction

Reaction

Reaction

16 cells + 40 cells → 40 cells

1. 16 cells
2. 40 cells
3. 40 cells



Pogil Introduction To Mole Concept

Alan Keith Griffiths



Pogil Introduction To Mole Concept:

Chemistry Education in the ICT Age Minu Gupta Bhowon, Sabina Jhaumeer-Laulloo, Henri Li Kam Wah, Ponnadurai Ramasami, 2009-07-21 th th The 20 International Conference on Chemical Education 20 ICCE which had rd th Chemistry in the ICT Age as the theme was held from 3 to 8 August 2008 at Le Meridien Hotel Pointe aux Piments in Mauritius With more than 200 participants from 40 countries the conference featured 140 oral and 50 poster presentations th Participants of the 20 ICCE were invited to submit full papers and the latter were subjected to peer review The selected accepted papers are collected in this book of proceedings This book of proceedings encloses 39 presentations covering topics ranging from fundamental to applied chemistry such as Arts and Chemistry Education Biochemistry and Biotechnology Chemical Education for Development Chemistry at Secondary Level Chemistry at Tertiary Level Chemistry Teacher Education Chemistry and Society Chemistry Olympiad Context Oriented Chemistry ICT and Chemistry Education Green Chemistry Micro Scale Chemistry Modern Technologies in Chemistry Education Network for Chemistry and Chemical Engineering Education Public Understanding of Chemistry Research in Chemistry Education and Science Education at Elementary Level We would like to thank those who submitted the full papers and the reviewers for their timely help in assessing the papers for publication th We would also like to pay a special tribute to all the sponsors of the 20 ICCE and in particular the Tertiary Education Commission <http://tec.intnet.mu> and the Organisation for the Prohibition of Chemical Weapons <http://www.opcw.org> for kindly agreeing to fund the publication of these proceedings

The Mole Concept Paul Gilchrist, 1982* *The Teaching of the Mole Concept* John Slater Allen, 1976 **The Mole Concept** Alan Keith Griffiths, 1979 *The Mole Concept*, 1983

SUMMARY A series of 6 programmes explaining the reasoning behind the development and acceptance of the mole concept

The Mole Concept : Investigation of an Hierarchical Model Alan K. (Alan Keith) Griffiths, 1979

Pogil Introduction To Mole Concept Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the power of words has become more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Pogil Introduction To Mole Concept**, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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