# 12.2 Chemical Calculations > Writing and Using Mole Ratios

$$N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$$

The three mole ratios derived from the balanced equation above are:

## Pogil Mole Ratios N2 G 3h2 G 2nh3 G

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### Pogil Mole Ratios N2 G 3h2 G 2nh3 G:

**Mole Ratios** Lifeliqe,2019 This lesson plan relates balanced chemical equations to everyday analogies defines stoichiometry and uses mole ratios to convert between amounts of substances in chemical reactions

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