

# Basic Solubility Rules (View 3)

General Solubility Rules (for water at 25°C)

| <u>Ion</u>                                                                                           | <u>Solubility</u> | <u>Exceptions</u>                                                                                                   |
|------------------------------------------------------------------------------------------------------|-------------------|---------------------------------------------------------------------------------------------------------------------|
| Alkali ions                                                                                          | soluble           | none                                                                                                                |
| $\text{NH}_4^{1+}$ , $\text{H}^{1+}$                                                                 | soluble           | none                                                                                                                |
| $\text{Cl}^{1-}$ , $\text{I}^{1-}$ , $\text{Br}^{1-}$                                                | soluble           | except $\text{Ag}^+$ , $\text{Hg}_2^{2+}$ , $\text{Pb}^{2+}$                                                        |
| $\text{CH}_3\text{COO}^{1-}$ ,<br>$\text{NO}_3^{1-}$                                                 | soluble           | none                                                                                                                |
| $\text{OH}^{1-}$                                                                                     | insoluble         | except Alkali ions, $\text{Ca}^{2+}$ , $\text{Ba}^{2+}$ , $\text{Sr}^{2+}$                                          |
| $\text{ClO}_4^{1-}$                                                                                  | soluble           | none                                                                                                                |
| $\text{SO}_4^{2-}$                                                                                   | soluble           | except $\text{Ca}^{2+}$ , $\text{Ba}^{2+}$ , $\text{Sr}^{2+}$ , $\text{Hg}^{2+}$ , $\text{Pb}^{2+}$ , $\text{Ag}^+$ |
| $\text{CO}_3^{2-}$ , $\text{PO}_4^{3-}$ ,<br>$\text{CrO}_4^{2-}$ , $\text{S}^{2-}$ , $\text{O}^{2-}$ | insoluble         | except Alkali ions and $\text{NH}_4^+$                                                                              |

Use this one – found on the back of our Periodic Table

# Pogil Solubility Rules Answers

**U. Bendict**



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**Solubility Rules Based on Lattice Parameter Differences** U. Bendict, 1977

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