6) What is the atomic radius of Carbon?

70pm

50 On each table, which element has the smallest atomic radius? What about the largest atomic radius?

Smallest H, N

Largest Rb, Ne



CY. 10) For the Group 1 table, what is the general frend of atomic tadks vs. position?

Atomic ractius increases as you go down

TD For the Period 2 bable, what is the general trend of storric radius in position?

Atomic radius decreases as you move across period



128 Are there any exceptions that you noticed for either of the trends?

Noble gases are significantly larger



139 Speculate with your group why this trend exception occurs. We will discuss as a class and why this exception occurs.

* Can discuss van der Waals radius V.S. Atomic radius & Nhy van der waals radius is used for noble gases

information. Generally, atomic radius increases as you go down and to the left on the periodic table.

As you go down on the periodic table, the energy level of the outer-most electron increases, which requires the electrons to be further from the nucleus.

As you go further to the right on the periodic table, the energy level of the outer-most electron doesn't change significantly. The effective nuclear charge does steadily increase though, which allows the nucleus to have a stronger force on the outer-most electrons of the atom.

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