

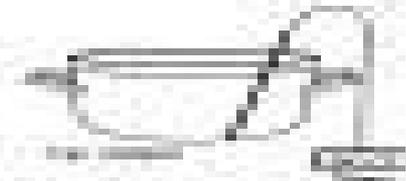
Calorimetry

What is calorimetry? How does it measure heat energy? How is it used?

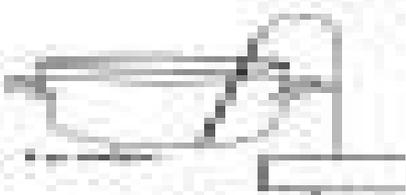
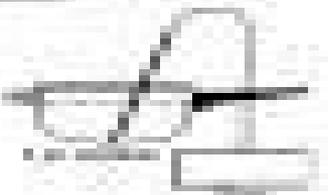
When a substance is heated, the temperature of that substance increases. Will the amount of energy cause different substances to have different temperature increases? Will the same amount of energy be needed to cause identical temperature increases in different amounts of the same substance? In this activity you will explore how mass, temperature, heat energy, and the type of substance are related.

Model 1 – A Hot Cup of Water

Activity



The energy to heat the water is provided by a hot object.



1. In Model 1, which substance had the most grams of water? _____

2. Complete the process described in Model 1.

a. How many total joules of energy were added to the beaker? _____

b. How many total joules of energy were added to the beaker? _____

3. In Model 1, which substance had the largest mass? Which substance had the smallest? Explain your answers.

4. How much heat energy was added to the water? _____

5. How much heat energy was added to the water? _____

Pogil Activities For High School Chemistry Calorimetry Packet

High School POGIL Initiative



Pogil Activities For High School Chemistry Calorimetry Packet:

POGIL Activities for High School Chemistry High School POGIL Initiative, 2012

CHEM-PACS, 1989

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