

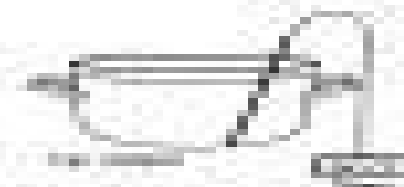
Calorimetry

What is calorimetry? How does it work? What is its purpose?

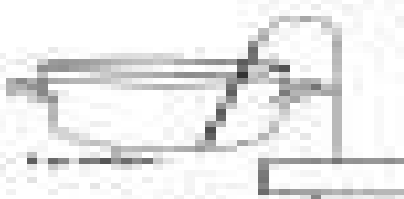
When a substance is heated, the temperature of that substance increases. Will the amount of energy (heat) required to heat different substances to the same temperature increase? Will the same amount of energy be needed to heat different temperatures? Increase or decrease amounts of the same substance to different temperatures will require less mass, temperature, heat energy, and the type of substance are related.

Model 1 - A Plot of Mass

Calorimetry



The energy (heat) required to heat a substance is related to its mass.



1. In Model 1, which substance (A or B) requires more energy to heat? Substance A

2. Compare the graphs (curves) in Model 1.

a. How many (what) units of energy were added to the substance? 100 J

b. How many (what) units of energy were added to the substance? 100 J

c. In Model 1, which substance (A or B) requires more energy to heat? Substance A

d. How many (what) units of energy were added to the substance? 100 J

Compare the graphs (curves) in Model 1. Substance A

Calorimetry

Pogil Activities For High School Chemistry Calorimetry Packet

High School POGIL Initiative



Pogil Activities For High School Chemistry Calorimetry Packet:

POGIL Activities for High School Chemistry High School POGIL Initiative, 2012

CHEM-PACS , 1989

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