

Radioactive Decay Equations POGIL

High School Nuclear Chemistry/Physics

Radioactive Decay Equations POGIL

Background: Radioactive Decay

Alpha Decay: This type of decay happens when a heavy nucleus (one having 82 or more protons) emits an α particle. An α particle is a helium nucleus, consisting of two protons and two neutrons.

For example, let's consider the decay of $^{238}_{92}\text{U}$. In this decay, the nucleus emits an α particle, which is a helium nucleus, and the resulting nucleus is $^{234}_{90}\text{Th}$.

The resulting alpha particle is a helium nucleus.



Gamma Decay: This decay happens when a nucleus in an excited state transitions to a lower energy state by emitting a gamma ray (γ).

For example, let's consider the decay of $^{60}_{27}\text{Co}$. In this decay, the nucleus emits a gamma ray, and the resulting nucleus is $^{60}_{27}\text{Co}$.

Decay Type	Resulting Nucleus
Alpha Decay	$^{234}_{90}\text{Th}$
Beta Decay	$^{234}_{91}\text{Pa}$
Gamma Decay	$^{60}_{27}\text{Co}$

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Background: Radioactive Decay

Beta Decay: This type of decay happens when a nucleus emits a beta particle. A beta particle is an electron (β^-) or a positron (β^+).

For example, let's consider the decay of $^{14}_6\text{C}$. In this decay, the nucleus emits a beta particle, and the resulting nucleus is $^{14}_7\text{N}$.

Positron Emission: This is a type of beta decay in which a nucleus emits a positron (β^+).

The resulting alpha particle is a helium nucleus.



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Gamma Decay: This decay happens when a nucleus in an excited state transitions to a lower energy state by emitting a gamma ray (γ).

For example, let's consider the decay of $^{60}_{27}\text{Co}$. In this decay, the nucleus emits a gamma ray, and the resulting nucleus is $^{60}_{27}\text{Co}$.

Pogil Activity For Nuclear Chemistry

Annelies Wilder-Smith



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Modern NMR Spectroscopy in Education David Rovnyak, 2007 This book is intended to be a comprehensive resource for educators seeking to enhance NMR enabled instruction in chemistry This book describes a host of new modern laboratories and experiments *Chemical Pedagogy* Keith S Taber, 2024-12-20 How should chemistry be taught in schools colleges and universities Chemical Pedagogy discusses teaching approaches and techniques the reasoning behind them and the evidence for their effectiveness The book surveys a wide range of different pedagogic strategies and tactics that have been recommended to better engage learners and provide more effective chemistry teaching These accounts are supported by an initial introduction to some key ideas and debates about pedagogy the science of teaching Chemical Pedagogy discusses how teaching innovations can be tested to inform research based practice Through this book the author explores the challenges of carrying out valid experimental studies in education and the impediments to generalising study results to diverse teaching and learning contexts As a result the author highlights both the need to read published studies critically and the value of teachers and lecturers testing out recommended innovations in their own classrooms Chemical Pedagogy introduces core principles from research into human cognition and learning to provide a theoretical perspective on how to

best teach for engagement and understanding An examination of some of the more contentious debates about pedagogy leads to the advice to seek optimally guided instruction which balances the challenge offered to learners with the level of support provided This provides a framework for discussing a wide range of teaching approaches and techniques that have been recommended to those teaching chemistry across educational levels including both those intended to replace teaching from the front and others that can be built into traditional lecture courses to enhance the learning experience *Making Chemistry Relevant* Sharmistha Basu-Dutt, 2010-02-19 Unique new approaches for making chemistry accessible to diverse students Students interest and achievement in academics improve dramatically when they make connections between what they are learning and the potential uses of that knowledge in the workplace and or in the world at large *Making Chemistry Relevant* presents a unique collection of strategies that have been used successfully in chemistry classrooms to create a learner sensitive environment that enhances academic achievement and social competence of students Rejecting rote memorization the book proposes a cognitive constructivist philosophy that casts the teacher as a facilitator helping students to construct solutions to problems Written by chemistry professors and research groups from a wide variety of colleges and universities the book offers a number of creative ways to make chemistry relevant to the student including Teaching science in the context of major life issues and STEM professions Relating chemistry to current events such as global warming pollution and terrorism Integrating science research into the undergraduate laboratory curriculum Enriching the learning experience for students with a variety of learning styles as well as accommodating the visually challenged students Using media hypermedia games and puzzles in the teaching of chemistry Both novice and experienced faculty alike will find valuable ideas ready to be applied and adapted to enhance the learning experience of all their students *An Introduction To Nuclear Chemistry* M. Satake, 2010 Contents The Development of Nuclear Chemistry Fundamental Particles and Nuclear Structure Radioactivity and Nuclear Reactions Properties of Nuclear Radiations The Detection and Measurement of Nuclear Radiation Nuclear Instrumentation Radiation Chemistry Isotope Measurement and Separation Methods Changed Particle Accelerators Neutron Sources Production and the Actinides Uses of Isotopes Experimental Nuclear Chemistry **Principles of Nuclear Chemistry** Russell R. Williams, 1950 **Experimental nuclear chemistry** Gregory R. Choppin, 1963

Nuclear Chemistry Maheshwar Sharon, Madhuri Sharon, 2009 Concentrating on techniques for the detection and measurement of radioactivity this book offers a guide to selecting the type of counter type of source sample duration for which the counting must be made and the radiation emitted by the isotope for its efficient detection It introduces a novel concept to explain not only the decay processes but also the selection of counting procedures for detecting and measuring radioactivity The author builds up the foundation from the nature of the interaction of radiation with matter He also highlights the differences between an ordinary chemical laboratory and a radiochemical one Provided by publisher

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