This is like a distribution in math. 2(X+3) -> 2X+6 Everything inside of the parenthets is multiplied by two.

6.	What does the subscripted ".	27	onwide the	parentheses of	the chemical	formula	sell y	post.	abou	150
	the compound?			State of the property		Agree 1		400		

The subscript "2" outside of the president indicate that there are two of the polyatomic ion nitrite (NOS).

20. How many atoms of		formula unit of ammonium	phosphate, (NH <sub>4</sub> ),PO <sub>4</sub>
nitrogen	hydrogen	phosphorus	coxygen

- A student writes the chemical formula for the ionic compound calcium hydroxide as CaOH.
  - at. Write the chemical formula for each ion in the compound. Calcium: Ca \*2 Hydroxide: OH
  - b. Why is the student's chemical formula for the compound calcium hydroxide wrong?

The subscript of "2" of the Avologes (H) indicate two Avologes, not true hydroxide polyatomic loss. Avolters should by placed a sum of the chemical formulas in Model 3 include parentheses. Which one of the following rules summarizes the appropriate use of parentheses in ternary ionic compounds? For the three rules

- that do not apply in all cases, show at least one counter example from the chemical formulas in (COH),
- Parentheses are used around any ion that is used more than once in a formula unit.

S Not true - Ala (CO3)3 - Al has more than one atom.

Parentheses are used around any polyatomic ion

Not true - KNO3 - NO3" has no potentiases because there is contyal

Parentheses are used around any polyatomic ion used more than once in a formula unit.

Parentheses are only used around polyatomic anions used more than once in a formula unit.

> Not true - (NH4), PO4 - ammonium's a polyatomic catlon.

Write chemical formulas for the following ternary ionic compounds.

G<sup>22</sup>SO<sub>y</sub><sup>-2</sup> → GSO<sub>y</sub> Cu<sup>2</sup>NO<sub>y</sub><sup>-1</sup> → Cu(NO<sub>y</sub>)<sub>2</sub> Li<sup>2</sup>NO<sub>y</sub><sup>-1</sup> → Li<sub>3</sub>PO<sub>y</sub>

d. Pocassium permanganate

e. Aluminum sulfite

f. Maemesium bianch

K" MaOy -> KMaOy A1 503 -> Al (503) 13 12 HOO3 -> Mg (HOO3) 2

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