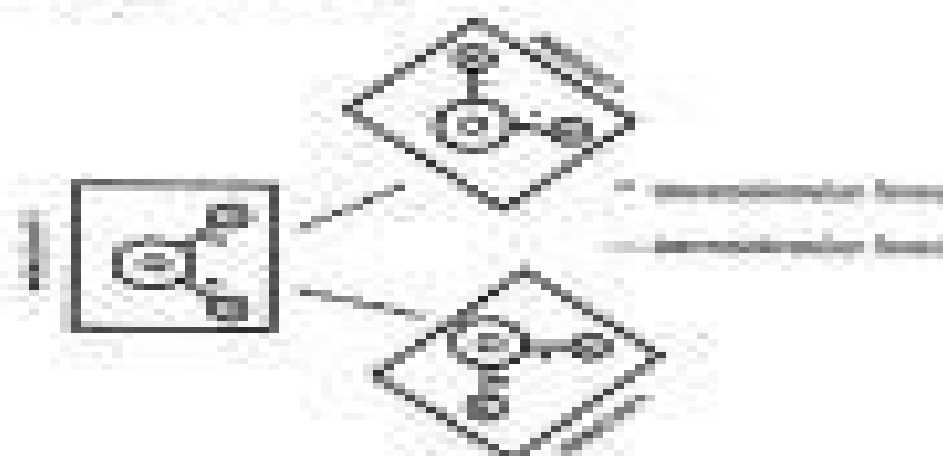


PHASE 1: Intermolecular Forces

[illegible]

The new "open" network, available to people up all diverse political belief systems, which characteristically continues to have weaknesses. It includes its own voice as independent media in that, groups of individuals "stand together" in order to form popular and public. The network that holds groups of individuals together are heterogeneous. Without representative media, the world we see today is not the same as the world we see today.

Figure 1. *Staphylococcus aureus* and *Enterococcus faecalis* strains.

1. Which specific molecule is represented inside each box in Figure 3?
2. In relation to the box for molecule 1, where do the intermolecular forces exist in Figure 1 — inside the boxes or outside the boxes?
3. Based on the intermolecular forces for molecule 1, how would molecule 1's boiling point compare to the boiling point of the molecule 2 and 3.
4. In relation to the molecule, where do intermolecular forces tend to occur — inside the molecule or outside of the molecule?
5. How intermolecular forces exist in Figure 1. Where are they positioned relative to the molecules — inside the molecules or between the molecules?

8. State the difference between intracellular and extracellular digestion of where they occur on the molecular level.
- Intracellular digestion – digestion that occurs within the cell.
- Extracellular digestion – digestion that occurs outside the cell.

Pogil Intermolecular Forces Answers

CO Houle



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Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes [Chemists' Guide to Effective Teaching](#) Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry *Intermolecular Forces* Geoffrey C. Maitland, 1981 [Intermolecular Forces](#) , [Intermolecular Forces](#) A. Pullman, 2014-01-15 **Theory of Intermolecular Forces** Henry Margenau, N. R. Kestner, 1969 **Intermolecular Forces** Taro Kihara, 1976 [Intermolecular Forces](#) Pierre L. Huyskens, Werner A. P. Luck, Thérèse Zeegers-Huyskens, 1991 The study of intermolecular forces began over one hundred years ago in 1873 with the famous thesis of van der Waals In recent decades knowledge of this field has expanded due to intensive research into both its theoretical and the experimental aspects This is particularly true for the type of very strong cohesive force stressed in 1920 by Latimer and Rodebush the hydrogen bond a phenomenon already outlined in 1912 by Moore and Winemill Hydrogen bonds exert a profound influence on most of the physical and chemical properties of the materials in which they are formed Not only do they govern viscosity and electrical conductivity they also intervene in the chemical reaction path which determines the kinetics of chemical processes The properties of chemical substances depend to a large extent on intermolecular forces In spite of this fundamental fact too little attention is given to these properties both in research and in university teaching For instance in the field of pharmaceutical research about 13000 compounds need to be studied in order to find a single new product that can be successfully marketed The recognition of the need to optimize industrial research efficiency has led to a growing interest in promoting the study of inter molecular forces Rising salary costs in industry have encouraged an interest in theoretical ideas which will lead to tailor made materials [Intermolecular Forces](#) A. Pullman, 1981-08-31 **Intermolecular Forces and Clusters I** , 2005

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