

Intermolecular Forces

Context: Properties of molecules are determined by both the force of the chemical bonds that hold molecules together and the forces that occur between separate molecules.

The forces between molecules are called **intermolecular forces** and are largely caused by the how the electrons in molecules are oriented relative to other molecules.

Because we cannot easily measure the exact strength of intermolecular forces, we will typically use boiling point and vapor pressure as stand-ins for the amount of intermolecular forces in a molecule.

Model 1: Table of Interactions Between Magnets and Metal

	Interaction	Factors
Magnet-Magnet		
Magnet-Metal		
Metal-Metal		

Questions:

1) For each arrangement of magnets and metal bring them together until they touch. Describe how the pair interact with each other as they approach each other and when they touch. Record these interactions on the table above.

2) In the magnet-magnet interaction, how did the poles of the magnet arrange themselves as the magnets approached each other?

Pogil Intermolecular Forces Ap

Ian Pickup



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Intermolecular Forces Pierre L. Huyskens, Werner A.P. Luck, Therese Zeegers-Huyskens, 2012-12-06 The study of intermolecular forces began over one hundred years ago in 1873 with the famous thesis of van der Waals. In recent decades knowledge of this field has expanded due to intensive research into both its theoretical and the experimental aspects. This is particularly true for the type of very strong cohesive force stressed in 1920 by Latimer and Rodebush, the hydrogen bond, a phenomenon already outlined in 1912 by Moore and Winemill. Hydrogen bonds exert a profound influence on most of the physical and chemical properties of the materials in which they are formed. Not only do they govern viscosity and electrical conductivity, they also intervene in the chemical reaction path which determines the kinetics of chemical processes. The properties of chemical substances depend to a large extent on intermolecular forces. In spite of this fundamental fact, too little attention is given to these properties both in research and in university teaching. For instance in the field of pharmaceutical research about 13000 compounds need to be studied in order to find a single new product that can be successfully marketed. The recognition of the need to optimize industrial research efficiency has led to a growing interest in promoting the study of intermolecular forces. Rising salary costs in industry have encouraged an interest in theoretical ideas which will lead to tailor made materials.

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