









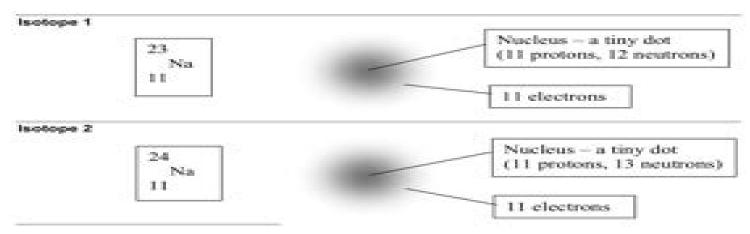
Atoms and their Inotopes

Subatomic Particles

Particle	Symbol	Relative Charge	Absolute Mass	Relative Mass
electron	360	-1	$9.109 \times 10^{-10} \text{ kg}$	0
proton	p+	+1	$1.673 \times 10^{-27} \text{ kg}$	1
neutron	n ^e	0	$1.675 \times 10^{-10} \text{ kg}$	1

Model: Two Isotopes of Sodium

The diagrams below show representations of sodium isotopes. [Note: the diameter of an atom is about 10,000 times larger than the diameter of the atomic nucleus so the relative sizes of the atom and the nucleus are not accurately depicted in these diagrams.]



Key Questions

What information is provided by the atomic number, Z?
The information powded by the atomic number(g) is
the number of priors and electors in a method atom.

What information is provided by the mass number, A?
the information pounded by the mass number (A) is the number of protons plus the number of necessary.

Pogil Isotopes Answer Key

Francis William Aston

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