

PHYSICAL KINETIC MOLECULAR THEORY

Learning Objectives

- 1) Identify theoretical differences in how gases participate in the kinetic theory and how molecules in liquids and solids participate in the kinetic molecular theory
- 2) Identify how temperature affects molecular motion
- 3) Explain how kinetic molecular theory is applied to the behavior of everyday substances

Module 1: Representation of States in Different Phases

The following figure illustrates representative particle behavior across different temperature regimes:



Figure 1: Represented states of matter at different temperature regimes

Molecular motion is different in different temperature regimes and phases. Kinetic molecular theory can help us understand why.

Key Concepts

1. Matter can be in three representative states and transitions in these states are called phase changes, and matter in these states has different motion of particles and different behavior (as shown in the figure). We can call this motion as kinetic. The amount of kinetic movement is called as the energy and not the velocity.

Table 1: Kinetic motion in the three states

	SOLID STATE	LIQUID STATE	GASEOUS STATE
How does the particles move?	Particles packed in a fixed, regular arrangement together. The motion is very low.	Particles are in a liquid, not regular, but they are close together. The motion is a regular, but not fixed.	Particles are in a gas, not regular, but they are far apart. The motion is a regular, but not fixed.
Are particles in this state in a fixed shape?	Yes, particles are in a fixed shape. They are packed together in a regular arrangement. The particles are not far apart, but they are close together.	No, particles are in a liquid. They are not packed together in a regular arrangement. The particles are not far apart, but they are close together.	No, particles are in a gas. They are not packed together in a regular arrangement. The particles are not far apart, but they are far apart.

Pogil Kinetic Molecular Theory Answers

Ilya Prigogine



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The Molecular Theory of Solutions Ilya Prigogine, 1957 The Molecular Theory of Solutions I. Prigogine, 2003-01
Molecules and the Molecular Theory of Matter Allan Douglas Risteen, 1895 **Molecular Theory of Solutions** Arie Ben-Naim, 2023 Based on the theory of Kirkwood and Buff this theory is a powerful and general tool to analyse study and understand any type of mixtures from the molecular or the microscopic point of view This book presents developments in the molecular theory of mixtures and solutions **Molecular Theory of Water and Aqueous Solutions** Arie Ben-Naim, 2009
The Potential Distribution Theorem and Models of Molecular Solutions Tom L. Beck, Michael E. Paulaitis, Lawrence R. Pratt, 2006-08-31 An understanding of statistical thermodynamic molecular theory is fundamental to the appreciation of molecular solutions This complex subject has been simplified by the authors with down to earth presentations of molecular theory Using the potential distribution theorem PDT as the basis the text provides a discussion of practical theories in conjunction with simulation results The authors discuss the field in a concise and simple manner illustrating the text with useful models of solution thermodynamics and numerous exercises Modern quasi chemical theories that permit statistical thermodynamic properties to be studied on the basis of electronic structure calculations are given extended development as is the testing of those theoretical results with ab initio molecular dynamics simulations The book is intended for students taking up research problems of molecular science in chemistry chemical engineering biochemistry pharmaceutical chemistry nanotechnology and biotechnology **Kinetic Molecular Theory** Charles Proteus Steinmetz, American Electrochemical Society. Meeting, 1908* **Molecular Theory of Water and Aqueous Solutions** Arie Ben-Naim, 2011 The aim of this book is to explain the unusual properties of both pure liquid water and simple aqueous solutions in terms of the properties of single molecules and interactions among small numbers of water molecules It is mostly the result of the author's own research spanning over 40 years in the field of aqueous solutions Jacket **Foundations of the Molecular Theory** Dalton, 1893 **Molecular Theory of Gases and Liquids** Joseph Oakland Hirschfelder, 2003-01 *The Molecular Theory of Gases and Liquids* Joseph O. Hirschfelder, Joseph Oakland Hirschfelder, Charles Francis Curtiss, Robert Byron Bird, 1954 An essential cross disciplinary reference for molecular interactions Molecular Theory of Gases and Liquids offers a rigorous comprehensive treatment of molecular characteristics and behaviors in the gaseous and fluid states A unique cross disciplinary approach provides useful insight for students of chemistry chemical engineering fluid dynamics and a variety of related fields with thorough derivations and in depth explanations throughout Appropriate for graduate students and working scientists alike this book details advanced concepts without sacrificing depth of coverage or technical detail *The Potential Distribution Theorem and Models of Molecular Solutions* Thomas L. Beck, Michael E. Paulaitis, Lawrence Riley Pratt, 2006 Modern quasi chemical theories that permit statistical thermodynamic properties to be studied on the basis of electronic structure calculations are given extended development as is the testing of those theoretical results with ab initio molecular

dynamics simulations The book is intended for students undertaking research problems of molecular science in chemistry chemical engineering biochemistry pharmaceutical chemistry nanotechnology and biotechnology **BOOK JACKET** **The Potential Distribution Theorem and Models of Molecular Solutions** Tom L. Beck, 2006 This book was first published in 2006 An understanding of statistical thermodynamic molecular theory is fundamental to the appreciation of molecular solutions This complex subject has been simplified by the authors with down to earth presentations of molecular theory Using the potential distribution theorem PDT as the basis the text provides a discussion of practical theories in conjunction with simulation results The authors discuss the field in a concise and simple manner illustrating the text with useful models of solution thermodynamics and numerous exercises Modern quasi chemical *Foundations of the Molecular Theory* John Dalton, Amedeo Avogadro, 2017-07-25 From the Preface THE papers here reprinted in chronological order serve to exhibit the historical development of the idea of a connection existing between the number of particles in different gases and the volume they occupy It will be seen that Dalton from the first entertains the notion that equal volumes of different gases may contain the same number of ultimate particles at equal temperature and pressure but that he is legitimately forced to reject this assumption conceiving no distinction between the atom and the molecule of an element Gay Lussac s important experimental work on the combining volumes of gases then shows the necessity of a simple relation between the ultimate particles of gases and their volumes although he does not point this out in his paper Dalton however perceives the necessity and characteristically concludes by doubting the accuracy of Gay Lussac s experiments Avogadro finally accepts both Dalton s theory and Gay Lussac s data and teaches how to reconcile them by distinguishing between the atom and the molecule of an elementary gas It has not been thought necessary to reprint the letter of Ampere to Berthollet Annales de Chimie 90 43 86 1814 since that paper contains no advance on the views of Avogadro published three years earlier its author simply drawing the same conclusions from the same premises The English version of the French originals will probably be found more faithful than elegant especially so in the case of Avogadro s paper where the French is always clumsy and occasionally obscure **MOLECULES AND THE MOLECULAR THEORY OF MATTER** A. D. RISTEEN, 2018 **Foundations of the Molecular Theory** John Dalton, 1899 **Foundations of the Molecular Theory** John Dalton, Joseph Louis Gay-Lussac, Amedeo Avogadro, 1893 **Foundations of the Molecular Theory Comprising Papers and Extracts** John Dalton, 1969 **Foundations of Molecular Theory, Comprising Papers and Extracts** John Dalton, Joseph Louis Gay-Lussac, Amedeo Avogadro, 1961 **The Kinetic Theory Half-elective** Alasdair John Hey, 1994-01-01

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