

Key

Molecular Geometry

How can molecular shapes be predicted using the VSEPR theory?

Why?

When you draw a Lewis structure for a molecule on paper, you are making a two-dimensional representation of the atoms. In reality however, molecules are not flat—they are *three-dimensional*. The true shape of a molecule is important because it determines many physical and chemical properties for the substance. In this activity you will learn how to predict molecular shapes.

Model 1 – Lewis Structures

Lewis Structures		3-D Molecular Shape
1. H_2CO	3 electron domains (3 bonding, 0 nonbonding)	trigonal planar
2. BeF_2	2 electron domains (2 bonding, 0 nonbonding)	Linear
3. CH_4	4 electron domains (4 bonding, 0 nonbonding)	Tetrahedral
4. NH_3	4 electron domains (3 bonding, 1 nonbonding)	pyramidal
5. H_2O	4 electron domains (2 bonding, 2 nonbonding)	Bent
6. CO_2	2 electron domains (2 bonding, 0 nonbonding)	linear

Lone pair = ..

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A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the Butterworth-Heinemann Limited text.

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Chemists' Guide to Effective Teaching Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry Molecular Geometry Alison Rodger, Mark Rodger, 2014-05-16 Molecular Geometry discusses topics relevant to the arrangement of atoms The book is comprised of seven chapters that tackle several areas of molecular geometry Chapter 1 reviews the definition and determination of molecular geometry while Chapter 2 discusses the unified view of stereochemistry and stereochemical changes Chapter 3 covers the geometry of molecules of second row atoms and Chapter 4 deals with the main group elements beyond the second row The book also talks about the complexes of transition metals and f block elements and then covers the organometallic compounds and transition metal clusters The last chapter tackles the consequences of small local variations in geometry The text will be of great use to chemists who primarily deal with the properties of molecules and atoms

Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL's theoretical basis its implementations in diverse environments and evaluation of student outcomes *The VSEPR Model of Molecular Geometry* Ronald J. Gillespie, Istvan Hargittai, 2012-01-01 Valence Shell Electron Pair Repulsion VSEPR theory is a simple technique for predicting the geometry of atomic centers in small molecules and molecular ions This authoritative reference written by the developer of VSEPR theory features extensive coverage of structural information as well as theory and applications Helpful data on molecular geometries bond lengths and bond angles appear in tables and other graphics 1991 edition Molecular Geometry Ronald James Gillespie, 1972 *The VSEPR Model of Molecular Geometry* Ronald James Gillespie, István Hargittai, 1991-01 A Pictorial Approach to Molecular Bonding and Vibrations John G. Verkade, 1997 Understanding molecular orbitals MOs is a prerequisite to appreciating many physical and chemical properties of matter This extensively revised second edition of A Pictorial Approach to Molecular Bonding presents the author's innovative approach to MOs generating them pictorially for a wide variety of molecular geometries A major enhancement to the second edition is the Pi and Macintosh compatible Nodegame software which is coordinated with the text and aids in pictorially teaching molecular orbital theory using generator orbitals *Physical Methods for Determining Molecular Geometry* Wallace S. Brey, 2000 Molecular Modelling and Bonding Elaine A. Moore, 2002 This Case Study including illustrations on CD ROM explores ways in which computer modelling in conjunction with experimental techniques is used to design new drugs **The**

Teaching of Molecular Geometry Vincenzo P. Civitella,1976 **The Shape and Structure of Molecules** Charles Alfred Coulson,1982 *Rodger Molecular Geometry* Butterworth-Heinemann Limited,1995-01-01 **Molecules and Models** Arne Haaland,2008-03-06 This book provides a systematic description of the molecular structures and bonding in simple compounds of the main group elements with particular emphasis on bond distances bond energies and coordination geometries The description includes the structures of hydrogen halogen and methyl derivatives of the elements in each group some of these molecules are ionic some polar covalent The survey of molecules whose structures conform to well established trends is followed by representative examples of molecules that do not conform We also describe electron donor acceptor and hydrogen bonded complexes Chemists use models to systematize our knowledge to memorize information and to predict the structures of compounds that have not yet been studied The book provides a lucid discussion of a number of models such as the Lewis electron pair bond and the VSEPR models the spherical and polarizable ion models and molecular orbital calculations and it outlines the successes and failures of each *Shape in Chemistry* Paul G. Mezey,1993 Molecular geometry Second Edition Gerardus Blokdyk,2018 Molecular geometry Second Edition **Physical Methods for Determining Molecular Geometry** Wallace Siegfried Brey (ifj),1965 **Molecular Geometry Optimization** Jenn-Tai Hwang,1973 **Lewis Structures and Molecular Geometry** Jeffrey R. Buell,1992 A Programmed Approach to Molecular Geometry George I. Sackheim, *Determination of Molecular Geometry by Spectroscopic Methods* ,1972

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