

## Chemistry pH and pOH Worksheet

1. Calculate the values of both pH and pOH of the following solutions.

	pH	pOH
0.020 M HCl		
0.0050 M NaOH		
A blood sample $7.2 \times 10^{-8}$ M of $H^+$		
0.0050 M NaOH		

2. Find the values of  $[H^+]$ , pOH, and  $[OH^-]$  that correspond to each of the following pH values.

	$[H^+]$	$[OH^-]$	pOH
pH of lemon juice = 2.9			
pH of sauerkraut = 3.85			
pH of milk of magnesia = 10.81			
pH of orange juice = 4.11			
pH of diluted household ammonia = 11.61			

3. Determine which of the solutions in problem #2 are acidic.
4. A certain brand of root beer has a hydrogen concentration equal to  $1.9 \times 10^{-5}$  M. What are the pH and pOH of this root beer?
5. Dr. Pepper has a  $[H^+] = 1.4 \times 10^{-5}$  M. What is its pH?

# Poh Practice Problems

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