

Chemical Equilibrium—Reversible Reactions

Refer to equilibrium equations in the discussion when answering these questions.

A. Saturated Sodium Chloride

1. What is the evidence for a shift in equilibrium? _____
2. Which ion caused the equilibrium to shift? _____
3. In which direction did the equilibrium shift? _____
4. If solid sodium hydroxide were added to neutralize the hydrochloric acid, would this reverse the reaction and cause the precipitated sodium chloride to redissolve? Explain.

B. Saturated Ammonium Chloride

1. What is the evidence for a shift in equilibrium? _____
2. In which direction did the equilibrium shift? _____
3. Which ion caused the equilibrium to shift? _____

C. Iron(III) Chloride plus Potassium Thiocyanate

1. What is the evidence for a shift in equilibrium when iron(III) chloride is added to the stock solution?
2. What is the evidence for a shift in equilibrium when potassium thiocyanate is added to the stock solution?

Report For Experiment 24 Reversible Reactions

D Kirk

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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