

Soil Test Interpretation Guide

D.A. Horneck, D.M. Sullivan, J.S. Owen, and J.M. Hart



EC 1478 • Revised July 2011

Regular soil testing is an important element in nutrient management. You can use soil tests as a diagnostic tool or to identify trends through time. To obtain meaningful test results, you must sample soil correctly, at the same time each year, and you must maintain records.

For more information, see the following in the OSU Extension Catalog (<http://extension.oregonstate.edu/catalog/>): *Soil Sampling for Home Gardens and Small Acreages* (EC 628); *Monitoring Soil Nutrients Using a Management Unit Approach* (PNW 570); *Monitoring Soil Nutrients in Dryland Systems Using Management Units* (EM 8920).

Soil testing laboratory methods vary, which may influence results and sufficiency ranges. Therefore, the sufficiency ranges in this publication are accurate only for the test methods listed.

Soil tests are used to measure soil nutrients that are expected to become plant-available. They do not measure the total amounts of nutrients in soil. Measurements of total nutrient content are not useful indicators of sufficiency for plant growth, because only a small portion of the nutrients are plant-available. Roots take up plant-available nutrients as positively or negatively charged ions from the soil (table 1).

Soil test results (see figure 1) can be viewed in three categories: (1) low or yes, a fertilizer addition *will* likely increase growth and yield; (2) high or no, a fertilizer addition *will not* likely increase growth or yield; and (3) intermediate or maybe, a fertilizer addition *may* increase growth or yield. Categorization of soil test results into “yes,” “no,” and “maybe” assists understanding the limits and benefits of using soil test results for making nutrient recommendations.

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Table 1. Plant-available nutrient forms.

Nutrient	Form used by plant
Cations (+):	
Nitrogen	NH_4^+
Potassium	K^+
Calcium	Ca^{2+}
Magnesium	Mg^{2+}
Manganese	Mn^{2+}
Copper	Cu^{2+}
Zinc	Zn^{2+}
Anions (–):	
Nitrogen	NO_3^-
Phosphorus	H_2PO_4^- and HPO_4^{2-}
Sulfur	SO_4^{2-}
Boron	H_2BO_3^- and H_3BO_3
Molybdenum	HMoO_4^- and MoO_4^{2-}
Chloride	Cl^-

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Kathleen Armour



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