

Year 10 Physics – Energy Revision Questions Part 2

Q1. A 'can-chiller' is used to make a can of drink colder.

The image below shows a can-chiller.



- (a) The initial temperature of the liquid in the can was 25.0°C .
The can-chiller decreased the temperature of the liquid to 20.0°C .
The amount of energy transferred from the liquid was 6530 J .
The mass of liquid in the can was 0.330 kg .

Calculate the specific heat capacity of the liquid.

Use the equation:

Energy transferred = mass $\times$ specific heat capacity $\times$ temperature change

Give the unit.

Specific heat capacity = _____ unit _____

(4)
(Total 4 marks)

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