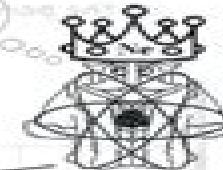


**Trends on the Periodic Table****Part 1: Identifying chemical properties/characteristics of the elements**

For each of the following elements, draw a caricature or cartoon above the description to show its "chemical personality" based on the description provided. (Note: See the Neon example for ideas.)

I want no positives... I have a full valence shell. I don't need others' electrons... I hope they just leave me alone!



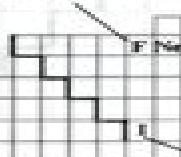
Neon  
Part of the Noble family, Neon does not readily interact with other elements. With a full set of valence electrons, Neon feels that he has everything he needs to make him happy without sharing with others.

**Lithium**  
Being one of the smaller elements with only 3 electrons, Lithium is happy to give one of its electrons away and never seems to ask for any in return.

Li → Lithium  
Lithium is a very nice element.



**Fluorine**  
One of the smallest elements, Fluorine is often described as sassy with a "magnetic" personality. As part of the halogen family fluorine likes to steal electrons from others. Fluorine is the best at convincing other elements to give up their electrons to her. However, she holds on tight to her own electrons and will not let them go without a fight.



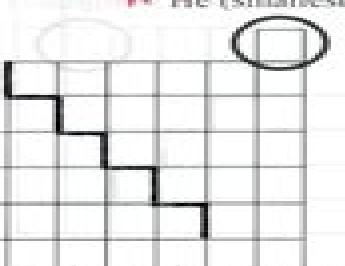
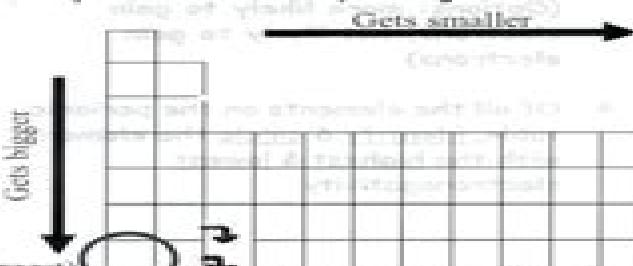
**Francium**  
Francium is the largest of all of the elements and has so many electrons that it has trouble keeping track of them. Francium will eagerly give up electrons to other elements and never takes in extras. But watch out, it has an extremely reactive and explosive personality!

Fr → Francium

**Part 2: Use your textbook to complete the following:****Atomic Radii Size:**

1. Explain **atomic radius**: A measure of the size (volume) of an atom. Determined by measuring half of the distance between two identical atoms bonded together.

2. Using the information in figure 5-13 (pg 141) in your book, write the atomic radii size for **alkali metals** & period #2 on the periodic table to your right.



3. What happens to atomic radii as you move down a column and across a row? On the arrows on the diagram, indicate the trend. (Options: gets bigger OR gets smaller)

4. Of all the elements on the periodic table, identify & circle the **largest** & **smallest** in size.

5. Rank the following elements from largest to smallest in atomic radii:

a) Be, Ba, Mg = Ba, Mg, Be

b) Li, F, B = Li, B, F

6. How can you explain the trend?

a. Why do they get smaller as you go across a period? Additional protons in the nucleus "pull" on electrons, bring the electron cloud closer. *Effective nuclear charge, Z<sub>eff</sub>, increases the attraction of the nucleus and therefore pulls the electron cloud closer to the nucleus resulting in smaller radius.*

Increases the attraction of the nucleus and therefore pulls the electron cloud closer to the nucleus resulting in smaller radius.

# Pogil Periodic Table Of The Elements Answer Key

**Adrian Dingle,Simon Basher**

## Pogil Periodic Table Of The Elements Answer Key:

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[The Basics of the Periodic Table](#) Leon Gray,2013-12-15 A sweeping history of both the discovery and classification of elements and the development of the modern periodic table Included are discussions of the discovery of

matter atoms atomic structure molecules compounds ions and isotopes as well as the first identifications of the 118 and counting elements and the various ways they have been classified and organized by prominent scientists up to the present day periodic table Instruction in how to read the periodic table is accompanied by examinations of the various groups of elements their location on the table and their properties and practical uses This text strongly supports Common Core Standards for the reading of scientific and technical texts and accounts and furnishes ample opportunities to summarize cite evidence and analyze connections between ideas individuals and events

**Elements and the Periodic Table, Grades 5 - 12**

Theodore A. Abbgy,2001-01-01 This informative classroom supplement is a great introduction to the periodic table explored in sequential form It includes activities transparency masters a teacher s guide an element game quizzes tests rubrics and answer keys Unit topics include discovering what elements are the uses of the elements element symbols periodic table organization and more Mark Twain Media Publishing Company specializes in providing captivating supplemental books and decorative resources to complement middle and upper grade classrooms Designed by leading educators the product line covers a range of subjects including mathematics sciences language arts social studies history government fine arts and character Mark Twain Media also provides innovative classroom solutions for bulletin boards and interactive whiteboards Since 1977 Mark Twain Media has remained a reliable source for a wide variety of engaging classroom resources

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Atoms, Molecules & Elements Gr. 5-8 George Graybill,2007-09-01 Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource makes the periodic table easier to understand Begin by answering what are atoms See how the atomic model is made up of electrons protons and neutrons Find out what a molecule is and how they differ from elements Then move on to compounds Find the elements that make up different compounds Get comfortable with the periodic table by recognizing each element as part of a group Examine how patterns in the period table dictate how those elements react with others Finally explore the three important kinds of elements metals nonmetals and inert gases Aligned to the Next Generation Science Standards and written to Bloom s Taxonomy and STEAM initiatives additional hands on experiments crossword word search comprehension quiz and answer key are also included

The Periodic Table Adrian Dingle,Simon Basher,2007 A mixture of

science and art creates a periodic table never seen before     [Periodic Table of the Elements](#) Ekkehard Fluck,Klaus G. Heumann,2003-10-17 Whenever you need to know something about a certain element keep a copy of this easy to use card in reach 3000 of the most important element data are assembled on this handy two sided reference card which is protected by a laboratory proof transparent cover The third edition incorporates the latest IUPAC recommendations including the official names for the elements 110 to 114 For each element the following data are available element symbol name and atomic number relative atomic mass isotope masses and natural abundance half life and predominant type of decay of radioactive species atomic ionic kovalent and van der Waals radius density melting point and boiling point electron configuration and electronegativity first ionization energy reduction potential oxidation states physiological significance     *Chemical Elements Pocket Guide* Coventry House Publishing,2020-02-08 The Chemical Elements Pocket Guide serves as a portable reference for quick study and efficient review of the 118 elements on the periodic table This on the go resource details the physical and atomic properties of each element as well as their history and characteristics in bullet point format The book s small trim size 4 25 x 6 8 inches is intended to fit inside a lab coat pocket and the bound design means you no longer need to carry loose bulky flashcards that can be misplaced or destroyed Includes the updated names nihonium moscovium tennessine and oganesson for elements 113 115 117 and 118 respectively Information provided includes Atomic number Atomic symbol Element category Standard state Atomic mass Electron configuration Oxidation states Electronegativity Atomic radius Ionization energy Electron affinity Melting point Boiling point Density Year discovered Discovered by Appearance Natural occurrence Interesting fact     **Periodic Table of the Elements** Ekkehard Fluck,Klaus G. Heumann,1986-07-09     *Quiz Yourself Clever! The Periodic Table* DK,2024-10-03 Get to grips with the periodic table with this fact packed quiz book Learn all the ingredients that make up the universe in this quiz book for children aged 9 that breaks down the periodic table Quiz Yourself Clever The Periodic Table goes through all 118 elements in an accessible and kid friendly way showing the pure form of every element through an eye catching image surrounded by quiz style questions This periodic table quiz book for children offers Fast and fun learning through the use of detailed images and interesting facts Hundreds of quick fire questions for children to quiz themselves and challenge friends and family Information that has been endorsed and authenticated by experts Vivid pictures and fast facts will allow children to gather a wealth of knowledge in a fun way Turn the page to find the answers contained in a handy dataset from its atomic number and mass to its melting point and when it was discovered Images of how it appears in nature and how we use it give a fuller understanding of each element More in the series If you enjoyed Quiz Yourself Clever The Periodic Table then why not test yourself other quiz books to boost your knowledge like Quiz Yourself Clever Rocks Minerals to get to grips with the fascinating rock and mineral forms all over the world or Animals of the World to learn about the most fascinating wildlife on our planet     **A Kids' Guide to the Periodic Table** Edward P. Zovinka,Rose A. Clark,2020-07-21 From aluminum to zinc make it fun for kids 8 to 12 to discover all 118

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**Leveled Texts: The Periodic Table** Joshua BishopRoby,2014-01-01

All students can learn about the periodic table through text written at four different reading levels Symbols on the pages represent reading level ranges to help differentiate instruction Provided comprehension questions complement the text

**The Periodic Table** Desmond Grosvenor Cooper,1964    *The Periodic Table* ,2008    *Periodic Table in Minutes* Dan Green,2016-05-19 An icon of science the Periodic Table defines the fundamental chemistry of everything in the universe In this compact yet comprehensive guide Dan Green outlines the history development and workings of the table shows how its design reflects and illuminates the organisation of all matter and even explains what it has to tell us about the chemistry of distant stars and of our own bodies Contents include an individual entry for every known element detailing properties uses and key data and sections on the patterns and groups of the famous table as well as explanations of basic chemistry concepts such as elements and compounds atomic structure chemical bonds reactions and radioactivity amongst many others

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## **Table of Contents Pogil Periodic Table Of The Elements Answer Key**

1. Understanding the eBook Pogil Periodic Table Of The Elements Answer Key
  - The Rise of Digital Reading Pogil Periodic Table Of The Elements Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Pogil Periodic Table Of The Elements Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Pogil Periodic Table Of The Elements Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Pogil Periodic Table Of The Elements Answer Key
  - Personalized Recommendations
  - Pogil Periodic Table Of The Elements Answer Key User Reviews and Ratings
  - Pogil Periodic Table Of The Elements Answer Key and Bestseller Lists

5. Accessing Pogil Periodic Table Of The Elements Answer Key Free and Paid eBooks
  - Pogil Periodic Table Of The Elements Answer Key Public Domain eBooks
  - Pogil Periodic Table Of The Elements Answer Key eBook Subscription Services
  - Pogil Periodic Table Of The Elements Answer Key Budget-Friendly Options
6. Navigating Pogil Periodic Table Of The Elements Answer Key eBook Formats
  - ePUB, PDF, MOBI, and More
  - Pogil Periodic Table Of The Elements Answer Key Compatibility with Devices
  - Pogil Periodic Table Of The Elements Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Pogil Periodic Table Of The Elements Answer Key
  - Highlighting and Note-Taking Pogil Periodic Table Of The Elements Answer Key
  - Interactive Elements Pogil Periodic Table Of The Elements Answer Key
8. Staying Engaged with Pogil Periodic Table Of The Elements Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Pogil Periodic Table Of The Elements Answer Key
9. Balancing eBooks and Physical Books Pogil Periodic Table Of The Elements Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Pogil Periodic Table Of The Elements Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Pogil Periodic Table Of The Elements Answer Key
  - Setting Reading Goals Pogil Periodic Table Of The Elements Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pogil Periodic Table Of The Elements Answer Key
  - Fact-Checking eBook Content of Pogil Periodic Table Of The Elements Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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