

Name: \_\_\_\_\_ Date: \_\_\_\_\_

### Periodic Table Scavenger Hunt

Search your periodic table for the answers to these questions.

1. Which element is number 53 on the periodic table?
2. What is the element symbol for gallium?
3. How many protons are in an atom of boron?
4. To which element group does argon belong?
5. Which element would you expect to have a higher mass: copper or iron?
6. What is the atomic mass of carbon?
7. What do you call the element series from atomic number 57-71?
8. Which element has a symbol that starts with a letter different from the first one in its name: aluminum, copper, gold, chromium?
9. Which element has the lowest atomic mass?
10. What is the first element with an atomic mass greater than 100?
11. What is the first basic metal on the periodic table?
12. True or false: The end antimony are in the same element group.
13. What is the heaviest alkali metal?
14. How many protons are in an atom of magnesium?
15. Which of the following is not a nonmetal: sulfur, oxygen, silicon, nitrogen?
16. What is the name of the element with the symbol Fe?
17. Which element has an atomic mass of 125.42?
18. Astatine belongs to which element group: nonmetal, halogen, noble gas?
19. What is the element with the symbol Ba?
20. Name a metal never used in any element symbol.

# fun PERIODIC TABLE SCAVENGER HUNT

# Periodic Table of the Elements

|                                                                                                                                                                                                                                                   |                          |                            |                          |                          |                         |                          |                          |                          |                          |                          |                           |                           |                           |                           |                           |                          |                           |                         |                         |                        |                        |                         |                         |
|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--------------------------|----------------------------|--------------------------|--------------------------|-------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|--------------------------|---------------------------|-------------------------|-------------------------|------------------------|------------------------|-------------------------|-------------------------|
| 1<br>1A<br>H<br>1.008                                                                                                                                                                                                                             | 2<br>2A<br>He<br>4.003   |                            |                          |                          |                         |                          |                          |                          |                          |                          |                           |                           |                           |                           |                           |                          |                           | 13<br>3A<br>B<br>10.81  | 14<br>4A<br>C<br>12.01  | 15<br>5A<br>N<br>14.01 | 16<br>6A<br>O<br>16.00 | 17<br>7A<br>F<br>18.99  | 18<br>8A<br>Ne<br>20.18 |
| 3<br>1A<br>Li<br>6.94                                                                                                                                                                                                                             | 4<br>2A<br>Be<br>9.01    |                            |                          |                          |                         |                          |                          |                          |                          |                          |                           |                           |                           |                           |                           |                          |                           | 19<br>3A<br>Al<br>26.98 | 20<br>4A<br>Si<br>28.09 | 21<br>5A<br>P<br>30.97 | 22<br>6A<br>S<br>32.07 | 23<br>7A<br>Cl<br>35.45 | 24<br>8A<br>Ar<br>39.95 |
| 5<br>1A<br>Na<br>22.99                                                                                                                                                                                                                            | 6<br>2A<br>Mg<br>24.31   | 7<br>3B<br>Sc<br>44.96     | 8<br>4B<br>Ti<br>47.88   | 9<br>5B<br>V<br>50.94    | 10<br>6B<br>Cr<br>52.00 | 11<br>7B<br>Mn<br>54.94  | 12<br>8B<br>Fe<br>55.85  | 13<br>8B<br>Co<br>58.93  | 14<br>8B<br>Ni<br>58.71  | 15<br>9B<br>Cu<br>63.55  | 16<br>10B<br>Zn<br>65.38  | 17<br>11B<br>Ga<br>69.72  | 18<br>12B<br>Ge<br>72.64  | 19<br>13B<br>As<br>74.92  | 20<br>14B<br>Se<br>78.96  | 21<br>15B<br>Br<br>79.90 | 22<br>16B<br>Kr<br>83.80  |                         |                         |                        |                        |                         |                         |
| 19<br>1A<br>K<br>39.10                                                                                                                                                                                                                            | 20<br>2A<br>Ca<br>40.08  | 21<br>3B<br>Sc<br>44.96    | 22<br>4B<br>Ti<br>47.88  | 23<br>5B<br>V<br>50.94   | 24<br>6B<br>Cr<br>52.00 | 25<br>7B<br>Mn<br>54.94  | 26<br>8B<br>Fe<br>55.85  | 27<br>8B<br>Co<br>58.93  | 28<br>8B<br>Ni<br>58.71  | 29<br>9B<br>Cu<br>63.55  | 30<br>10B<br>Zn<br>65.38  | 31<br>11B<br>Ga<br>69.72  | 32<br>12B<br>Ge<br>72.64  | 33<br>13B<br>As<br>74.92  | 34<br>14B<br>Se<br>78.96  | 35<br>15B<br>Br<br>79.90 | 36<br>16B<br>Kr<br>83.80  |                         |                         |                        |                        |                         |                         |
| 37<br>1A<br>Rb<br>85.47                                                                                                                                                                                                                           | 38<br>2A<br>Sr<br>87.62  | 39<br>3B<br>Y<br>88.91     | 40<br>4B<br>Zr<br>91.22  | 41<br>5B<br>Nb<br>92.91  | 42<br>6B<br>Mo<br>95.94 | 43<br>7B<br>Tc<br>98.91  | 44<br>8B<br>Ru<br>101.07 | 45<br>8B<br>Rh<br>102.91 | 46<br>8B<br>Pd<br>106.42 | 47<br>9B<br>Ag<br>107.87 | 48<br>10B<br>Cd<br>112.41 | 49<br>11B<br>In<br>114.82 | 50<br>12B<br>Sn<br>118.71 | 51<br>13B<br>Sb<br>121.76 | 52<br>14B<br>Te<br>127.60 | 53<br>15B<br>I<br>126.91 | 54<br>16B<br>Xe<br>131.29 |                         |                         |                        |                        |                         |                         |
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# Reading The Periodic Table Scavenger Hunt Answer

**Tom Jackson**



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**The Periodic Table: Book and Fact Cards** Janet Bingham,2025-02-04 Provides a wonderful resource for learning and revision The flashcards are a great way for children to test their knowledge

*The Periodic Table of Elements* Jenny Karpelenia,2006-09-01 Jordan skips school and heads To The mountains with his best friend Lee Their carefree trip turns serious when Jordan s car dies in the midst of a blizzard Passages to Adventure Hi Lo Novel

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