

4A-1

1. la biblioteca
2. el gimnasio
3. la lección de piano
4. las montañas (al campo)
5. al cine
6. la playa
7. el parque
8. restaurante

4A-2

Answers will vary, but should be structured as follows:

1. Voy al trabajo para trabajar.
2. Voy a la biblioteca para leer y estudiar.
3. Voy a la clase de español para hablar español.
4. Voy al parque para comer y caminar.
5. Voy al centro comercial para ir de compras.
6. Voy a la lección de piano para tocar el piano.
7. Voy al restaurante para comer y beber.
8. Voy al cine para ver una película.
9. Voy a la piscina para nadar.
10. Voy al gimnasio para hacer ejercicio.
11. Voy al centro comercial para estar con amigos.
12. Voy al gimnasio para levantar pesos.

4A-3

1. —estudias
—Voy a la biblioteca
2. —corres
—Voy al parque.
3. —levantas pesos
—Voy al gimnasio
4. —caminas
—Voy a las montañas.
5. —vas de compras
—Voy al centro comercial
6. —nadas
—Voy a la piscina

4A-4

1. Tito va al cine los viernes.
2. Tito estudia en la biblioteca los miércoles.
3. Tito hace ejercicio los sábados (y los martes).

4. Generalmente, Tito trabaja los lunes y los jueves.
5. Tito trabaja los lunes.
6. Tito va a la iglesia los domingos.
7. Tito va al gimnasio y a la iglesia los fines de semana.

4A-5

1. LOUIS: voy
ELIA: vamos
LOUIS: va
2. MARTA: vas
JUAN: voy / voy
MARTA: van / voy
JUAN: van
MARTA: vamos
3. RODOLFO: _____
PASO Y FELICE: _____
RODOLFO: van
PASO: voy
FELICE: voy / Voy
RODOLFO: vamos / va
PASO: va / van
FELICE: ir

4A-6

A.

1. Quién
2. Cuándo
- 3.Cuál
4. Cómo
5. Adónde
6. Dónde
7. Qué
8. Cuántos

B.

Answers will vary. Do not repeat questions from part A.

4A-7

Answers may vary.

- Tus padres: ¿Adónde van esta noche?
Tus padres: ¿Con quiénes van?
Tus padres: ¿Cómo se llaman ellos?
Tus padres: ¿De dónde son?
Tus padres: ¿Cómo es Roberto?
Tus padres: ¿Y cómo es Ana?
Tus padres: ¿Qué van a hacer después de ir al restaurante?
Tus padres: ¿Y después (de eso)?

Crucigrama (4A-8)

Across:

1. templo
4. campo
8. digas

9. merquita
11. quién
12. lección
14. libre
15. compras
16. después
18. película
19. montañas

Down:

2. playa
3. iglesia
5. parque
6. biblioteca
7. centro
10. gimnasio
13. cine
17. piscina
20. trabajo

Organizer (4A-9)

I. Vocabulary: Answers will vary.

II. Grammar

- | | |
|------------|---------|
| 1. col. 1. | col. 2. |
| voy | vamos |
| vas | vais |
| va | van |
2. A. quién, qué, dónde, cuándo,
por qué, cómo
B. before

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E Durkheim



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ESL and Spanish History Workbook Intermediate and Advanced Level Vocabulary Games for the Effective Multicultural Classroom Professor Snyder, 2020-02-27 The ESL and Spanish History Workbook is a high interest multicultural student activity vocabulary development workbook with an intermediate ESL curriculum guide ESL lesson plans and a teacher's model A complete answer key to the student activities is included to facilitate teacher time management The book was designed for busy language teachers and parents This workbook can be used as an independent study ESL book The ESL and Spanish History Workbook has several vocabulary sentence structure mapping and reading comprehension word games designed for intermediate to advanced level ESL students and or for students studying Spanish history The book's focus is on Spanish history vocabulary and reading development as well as multicultural education appreciation The book was designed to motivate secondary and college level ESL students and or anyone studying Spanish history It was designed to supplement language learning The word games can be used to review reinforce and celebrate Spanish history The vocabulary games are appropriate for individual as well as group work The exercises can be used in class or as brief take home assignments The activities were designed to be enjoyable challenging and most importantly enlightening They were calculated to save teachers time by supplying ready to use supplementary material Vocabulary is presented in context to the theme of Spanish history in an effort to increase knowledge and understanding of the diverse cultures of the people of the United States Students will increase their reading writing and vocabulary skills through exposure to Spanish history customs traditions institutions and values Language teachers will develop a better understanding of ESL intermediate level goals objectives and instructional planning Enjoy the book

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you look at a chemical equation for a reaction when you see a chemical equation in a textbook it looks like the reactants are completely used up and all made into product

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web jun 2 2020 the reactant that is consumed first and limits the amount of product s that can be obtained is the limiting reactant to identify the limiting reactant calculate the number of moles of each reactant present and compare this ratio to the mole ratio of the reactants in the balanced chemical equation

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web a which chemical is the limiting reactant b how many grams of zns will be formed c how many grams of the excess reactant will remain after the reaction is over 3 which element is in excess when 3 00 grams of mg is ignited in 2 20 grams of pure oxygen what mass is in excess what mass of mgo is formed 4

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web reactant present in an amount lower than required by the reaction stoichiometry thus limiting the amount of product generated molecular equation chemical equation in which all reactants and products are represented as neutral substances

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web the key to recognizing which reactant is the limiting reagent is based on a mole mass or mass mass calculation whichever reactant gives the lesser amount of product is the limiting reagent what we need to do is determine an amount of

one product either moles or mass assuming all of each reactant reacts

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web the reactant which is in a lesser amount than is required by stoichiometry is the limiting reactant in an alternate method of finding the limiting reagent the amount of product formed by each reactant is calculated the limiting reactant is the reactant from which the minimum amount of product is formed also if we calculate the amount of

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web the limiting reactant or limiting reagent is the reactant that gets consumed first in a chemical reaction and therefore limits how much product can be formed as we saw in example 1 there are many different ways to determine the limiting reactant but they all involve using mole ratios from the balanced chemical equation

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