

- 11) All the other substances besides SO_2 , HCl , & NH_3 are ionic compounds which means they are what physical state?
solids
- 12) What conclusions can you draw about the effects of temperature on the solubility for gases compared to that of solids?
As temperature increases, the solubility of gases decreases but the solubility of solids increases.
- 13) What substances have saturation points that are the most vulnerable to fluctuations in temperature?
 KNO_3 , NaNO_3 & NH_3
- 14) What substance has a saturation point that is most resistant to changes in temperature?
 NaCl
- 15) Suppose 12 grams of SO_2 is dissolved in water at 1°C . The water is then heated. At what temperature will the SO_2 bubble out of the solution?
About 12°C
- 16) Suppose 40 grams of KClO_3 is dissolved in water at 95°C . The water is then cooled. At what temperature will the KClO_3 precipitate out of the solution?
About 78°C
- 17) Suppose 40 grams of NH_4Cl is dissolved in water at 85°C . The water is then cooled to 50°C . Will a precipitate form? Explain your answer.
No, the saturation point of NH_4Cl at 50°C is about 51 grams so no solid precipitate will form.

- 18) 90 grams of KNO_3 is placed in 100ml of water @ 40°C ...
- a) Will the solution be saturated? If the answer is yes, how much (if any) KNO_3 will remain undissolved at the bottom of the container? If the answer is no, how much more KNO_3 could be dissolved?
Yes, about 28 grams will remain undissolved.
- b) If the solution is heated to 70°C will it be saturated? If the answer is yes, how much (if any) KNO_3 will be undissolved at the bottom of the container? If the answer is no, how much more KNO_3 could be dissolved?
No, about 45 more grams of KNO_3 could be dissolved at this temperature
- c) If the solution is cooled to 20°C will it be saturated? If the answer is yes, how much (if any) KNO_3 will be undissolved at the bottom of the container? If the answer is no, how much more KNO_3 could be dissolved?
Yes, about 58 grams will be re-crystallized or undissolved.
- 19) 90 grams of NaNO_3 is placed in 100ml of water @ 40°C ...
- a) Will the solution be saturated? If the answer is yes, how much (if any) NaNO_3 will remain undissolved at the bottom of the container? If the answer is no, how much more NaNO_3 could be dissolved?
No, about 15 more grams could be dissolved.
- b) If the solution is heated to 70°C will it be saturated? If the answer is yes, how much (if any) NaNO_3 will be undissolved at the bottom of the container? If the answer is no, how much more NaNO_3 could be dissolved?
No, about 45 more grams of NaNO_3 could be dissolved
- c) If the solution is cooled to 20°C will it be saturated? If the answer is yes, how much (if any) NaNO_3 will be undissolved at the bottom of the container? If the answer is no, how much more NaNO_3 could be dissolved?
Yes, about 3 grams will be re-crystallized.

Pogil Solubility Curve Answer Key

Liyong Dong

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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