

# Oxidation and Reduction

When happens when a substance is oxidized or a chemical reaction?

## Q1) Eg1

Below reactions show the process of corrosion with either magnesium or zinc as a copper gets oxidized in a chemical reaction process as it reacts. Metals react or corrode in the presence of an acid water. Sometimes these reactions are coupled to reactions for decomposition with the use of electricity as possible from metals and nonmetals. All of these reactions are examples of oxidation and reduction, otherwise known as redox reactions. In this activity you will explore what is happening on the atomic level in redox reactions.

## Model 1 – Redox Reactions

### Redox Reactions



### Redox Reactions



1. What are typical reactions in Model 1?

Redox reactions and chemical reactions.

2. Describe the atoms and molecules involved in Model 1. What does each represent in the redox reactions? How would you describe the atoms and molecules? Do you see any specific strengths from Model 1 as representing redox reactions?

There is no obvious charge transfer between different substances.

# Pogil Chemistry Answer Key Oxidation

**American Chemical Society, Milos  
Hudlicky**



## **Pogil Chemistry Answer Key Oxidation:**

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Oxidation and Reduction Brian J. Knapp, 1998 Theories and concepts of oxidation and reduction reactions are explained through the use of a series of illustrated step by step demonstrations

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Oxidation and Reduction in Inorganic and Analytical Chemistry Alan Vincent, Principles of Oxidation and Reduction Alan George Sharpe, 1965

**A Study of Reversible Oxidation and Reduction Reactions in Solutions** Eugene Paul Schoch, 1904

**Oxidation Mechanisms** Ross Stewart, 1964

**PRINCIPLES OF OXIDATION AND REDUCTION** Alan G. Sharpe, 1969

**Oxidation and Reduction in Organic Chemistry from the Standpoint of Potential Differences; The System Hydroquinone and Quinone** Frederic Stearns Granger, 2018-10-13 This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work is in the public domain in the United States of America and possibly other nations Within the United States you may freely copy and distribute this work as no entity individual or corporate has a copyright on the body of the work Scholars believe and we concur that this work is important enough to be preserved reproduced and made generally available to the public To ensure a quality reading experience this work has been proofread and republished using a format that seamlessly blends the original graphical elements with text in an easy to read typeface We appreciate your support of the preservation process and thank you for being an important part of keeping this knowledge alive and relevant

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Hudlicky,1996-01-01     **Oxidation** Barry M. Trost,1991 Volume 7     *Oxidation-reduction and Descriptive Chemistry*  
Charles E. Ophardt,1982     *Oxidation-reduction Reactions* Irish Science Teachers' Association. Chemistry Syllabus  
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