

Reaction Rates Test #11

Name: _____

____K ____T ____C ____A
(Product)

Assessment of Learning

10 10 10 10

Multiple Choice: Circle the most correct answer. (10K)

- Which of the following statements is **not** generally correct?
 A) endothermic reactions have larger energies of activation than exothermic reactions
 B) the rate of a reaction is proportional to its activation energy
 C) the rate of a reaction generally increases as the temperature is raised
 D) all reactions in which bonds are broken and formed have a significant activation energy

- Consider the following reaction:



The rate of this reaction can be expressed as...

- $\text{rate} = [\text{O}_2]$
- $\text{rate} = \frac{\Delta[\text{O}_2]}{\Delta t}$
- $\text{rate} = \frac{\Delta[\text{Hg}]}{\Delta t}$
- $\text{rate} = \frac{\Delta[\text{HgO}]}{\Delta t}$

- Which of the following could be used to describe the rate of a reaction?

- | | |
|-------------------------------------------------------------------|-------------------------------------------------------------------|
| A) $\frac{\text{change in time}}{\text{change in concentration}}$ | B) $\frac{\text{change in mass}}{\text{change in concentration}}$ |
| C) $\frac{\text{change in concentration}}{\text{change in time}}$ | D) $\frac{\text{change in concentration}}{\text{change in mass}}$ |

- Which of the following changes occur when the temperature of a reaction is increased?

- ΔH of the reaction increases
- Frequency of the collisions increases

Reaction Rates Assessment Answers

Brittany Deaton

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the name bar.

Reaction Rates Assessment Answers:

Digest of Education Statistics, 2000 **Science Educator's Guide to Laboratory Assessment** Rodney L. Doran, 2002
The book opens with an up to date discussion of assessment theory research and uses Then comes a wealth of sample assessment activities in biology chemistry physics and Earth science Keyed to the National Science Education Standards the activities include reproducible task sheets and scoring rubrics All are ideal for helping students reflect on their own learning during science lab *Proceedings*, 1992 *Serviceology for Designing the Future* Takashi Maeno, Yuriko Sawatani, Tatsunori Hara, 2016-09-18 This book provides a useful overall guide to the state of the art in theory and practice of services It can also serve as a reference book for researchers in various fields including engineering marketing economics and other disciplines Advanced works presented here were selected from the proceedings of the Second International Conference on Serviceology ICServ2014 held September 14 16 2014 This book helps readers to understand serviceology which tackles with a broad range of services the globalization of the economy and also enhances the quality of life of local residents *The Condition of Education*, 1995 **Chemistry3** Andrew Burrows, John Holman, Andrew Parsons, Gwen Pilling, Gareth Price, 2017 Chemistry3 establishes the fundamental principles of all three strands of chemistry organic inorganic and physical By building on what students have learned at school using carefully worded explanations annotated diagrams and worked examples it presents an approachable introduction to chemistry and its relevance to everyday life
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Development of Survey Methods to Assess Survey Practices Barbara Ann Bailar,C. Michael Lanphier,1978

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