

CHAPTER 1.3

Rates of Reaction

■ SOLUTIONS TO EXERCISES

Acetaminophen tablets that contain 500 mg of the compound, $\text{C}_8\text{H}_9\text{NO}_2$, are sometimes found with the acetaminophen content that has slight variation in content. The high content of the compound, the lowest measured acetaminophen content, or both may indicate adulteration of the product with non-acetaminophen based material; only the first one and the first one only.

110. Rate of formation of H_2S = $\frac{\text{change in } [\text{H}_2\text{S}]}{\text{time}}$. Rate of loss of H_2S = $-\frac{\text{change in } [\text{H}_2\text{S}]}{\text{time}}$. Rate of loss of H_2S is $\frac{\text{change in } [\text{H}_2\text{S}]}{\text{time}}$.

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111. Rate = $\frac{\text{change in } [\text{H}_2\text{S}]}{\text{time}}$. Rate of loss of H_2S = $-\frac{\text{change in } [\text{H}_2\text{S}]}{\text{time}}$. Rate of loss of H_2S = $-\frac{\text{change in } [\text{H}_2\text{S}]}{\text{time}}$.

112. The rate of loss of H_2S is $-\frac{\text{change in } [\text{H}_2\text{S}]}{\text{time}}$. The rate of loss of H_2S is $-\frac{\text{change in } [\text{H}_2\text{S}]}{\text{time}}$.

113. The rate of loss of H_2S is $-\frac{\text{change in } [\text{H}_2\text{S}]}{\text{time}}$. The rate of loss of H_2S is $-\frac{\text{change in } [\text{H}_2\text{S}]}{\text{time}}$.

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CH Cherryholmes



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