

Types of Chemical Reactions Lab Answer Key

Chemical Reaction #1

<i>Reactants (write in words)</i>	hydrochloric acid and magnesium
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	single replacement
<i>Predict Products (write in words)</i>	magnesium chloride and hydrogen gas
<i>Equation (write in symbols)</i>	$2\text{HCl}_{\text{(aq)}} + \text{Mg}_{\text{(s)}} \rightarrow \text{MgCl}_{2\text{(aq)}} + \text{H}_{2\text{(g)}}$

Chemical Reaction #2

<i>Reactants (write in words)</i>	copper metal and oxygen gas
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	synthesis
<i>Predict Products (write in words)</i>	copper oxide
<i>Equation (write in symbols)</i>	$2\text{Cu}_{\text{(s)}} + \text{O}_{2\text{(g)}} \rightarrow 2\text{CuO}_{\text{(s)}}$

Chemical Reaction #3

<i>Reactants (write in words)</i>	methane gas and oxygen gas
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	combustion
<i>Predict Products (write in words)</i>	carbon dioxide and water
<i>Equation (write in symbols)</i>	$\text{CH}_{4\text{(g)}} + 2\text{O}_{2\text{(g)}} \rightarrow \text{CO}_{2\text{(g)}} + 2\text{H}_2\text{O}_{\text{(g)}}$

Chemical Reaction #4

<i>Reactants (write in words)</i>	hydrogen peroxide
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	decomposition
<i>Predict Products (write in words)</i>	water and oxygen gas
<i>Equation (write in symbols)</i>	$2\text{H}_2\text{O}_{2\text{(l)}} \rightarrow 2\text{H}_2\text{O}_{\text{(l)}} + \text{O}_{2\text{(g)}}$

Reaction Lab Answer Key Hydrocarbon

ML Morrison

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Reaction Lab Answer Key Hydrocarbon:

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[Synthesis of Hydrocarbons in the Gasoline Range, 1938-1953](#) American Petroleum Institute. Research Project 45,1953

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From Source to Seep M. Lawson, M.J. Formolo, J.M. Eiler, 2018-03-28

Hydrocarbon systems by nature are a complex interplay of elements that must be spatially and temporally aligned to result in the generation and preservation of subsurface hydrocarbon accumulations. To meet the increasing challenges of discovering hydrocarbon resources, it is essential that we advance our understanding of these systems through new geochemical approaches and analytical developments. Such development requires that academic and industry-led research efforts converge in ways that are unique to the geosciences. The aim of this volume is to bring together a multidisciplinary geochemical community from industry and academia working in hydrocarbon systems to publish recent advances and state-of-the-art approaches to resolve the many remaining questions in hydrocarbon systems analysis. **From Source to Seep** presents geochemical and isotopic studies that are grouped into three themes: 1) source rock identification and the temperature timing of hydrocarbon generation; 2) mechanisms and time scales associated with hydrocarbon migration, trapping, storage, and alteration; and 3) the impact of fluid flow on reservoir properties.

Journal of the Institution of Petroleum

Technologists, 1920. **Journal of the Institution of Petroleum Technologists**. Institute of Petroleum (Great Britain), 1920. Vols for 19 include abstracts.

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