#### Equilibrium Practice Problems Worksheet

- In the reaction 2 NO₂(g) ↔ N₂O₄(g) the initial concentration of NO₂ was 0.250 M and N₂O₄ was 0.000 M. At equilibrium, the concentration of N₂O₄ was measured as 0.0133M.
  - a. Calculate the equilibrium concentration of NO<sub>2</sub>
  - b. Calculate K<sub>c</sub>.

	NO <sub>2</sub>	N <sub>2</sub> O <sub>4</sub>
Initial	0.250	0.000
Change	-2x	+x
Equilibrium	0.250 - 2x	0.0133

We can see from the last column that x must be 0.0133 M.

- (a) We can now calculate the equilibrium value of  $NO_2$  as 0.250-2x=0.250-0.0266=0.2234 M
- (b) Hence the value for K, is

$$K_c = \frac{[N_2O_4]}{[NO_2]^2} = \frac{0.0133 \text{ M}}{0.2234 \text{ M}^2} = 0.27$$

 0.05 mol Hz(g) and 0.05 mol Brz(g) are placed together in a 5.0 L flask and heated to 700 K. What is the concentration of each substance in the flask at equilibrium if Kz = 64 at 700 K?

$$H_2(g) + Br_2(g) \leftrightarrow 2 HBr(g)$$

The initial concentrations are  $[H_2] = [Br_2] = 0.05 \text{ mol} / 5.0 \text{ L} = 0.01 \text{ M}$  and [HBr] = 0.

	H <sub>2</sub> (g)	$Br_2(g)$	HBr(u)
Initial	0.01 M	0.01 M	0
Change	- X	- X	+ 2x
Equilibrium	0.01 - x	0.01 - x	2x

\*Using the hundred rule says that we cannot ignore the x in the [Hz] and [Brz]

# Reaction Rates And Equilivrium Practice Problems Answer

Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk **Reaction Rates And Equilivrium Practice Problems Answer:** 

# Reaction Rates And Equilivrium Practice Problems Answer Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Reaction Rates And Equilivrium Practice Problems Answer**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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