

Types of Chemical Reaction and Balancing Guide (CEI practice)

Identify

- Decide the correct symbol for the type of reaction
- Place the correct in the blanks at the beginning of each equation and
- Balance each equation correctly

- A is synthesis B is single displacement
 C is decomposition D is double displacement
 E is combustion

1. _____ $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$

2. _____ $\text{Fe} + \text{S} \rightarrow \text{FeS}$ $\text{Fe} + \text{S} \rightarrow \text{FeS}$ $\text{Fe} + \text{S} \rightarrow \text{FeS}$ $\text{Fe} + \text{S} \rightarrow \text{FeS}$

3. _____ $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$ $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$ $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$ $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$

4. _____ $\text{Fe} + \text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2$ $\text{Fe} + \text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2$ $\text{Fe} + \text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2$ $\text{Fe} + \text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2$

5. _____ $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$

6. _____ $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$

7. _____ $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$ $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$ $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$

8. _____ $\text{Fe} + \text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2$ $\text{Fe} + \text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2$ $\text{Fe} + \text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2$ $\text{Fe} + \text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2$

9. Balance the given equation and identify the reaction type. Write the balanced equation in the space provided. Identify the reaction type.

Reaction Types Quiz B Answers

Wolfgang Guggemos

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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