

Organic Chemistry II Study Guide for Exam I  
Chapters 10, 11, and 14

- Classification of Alcohols
  - Alcohols are classified according to how many carbons are attached to the carbon containing the -OH group.

- Methyl, primary, secondary, and tertiary

- Nomenclature of Alcohols

- Naming an alcohol

- Start numbering the carbon chain to reach the -OH group first
    - Include the location (carbon number) of the -OH group



- 2-methylhexan-2-ol

- Naming an alcohol as a substituent

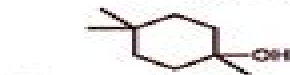
- Do this when alcohol is NOT the highest priority group (carboxylic acid, etc.)
    - Start numbering the carbon chain at the higher priority group
    - Use the prefix hydroxy- and include the location (carbon number) of the -OH group



- 4-hydroxybutanoic acid

- Naming cyclic alcohols

- If the -OH is the highest priority group, do NOT include the location (carbon number)
    - It is implied so it is unnecessary



- 1,4,4-trimethylcyclohexanol

- Naming diols

- When 2 -OH groups are present and is the highest priority group, use the suffix -diol
    - Include the location (carbon number) of BOTH -OH groups



- Propane-1,3-diol

# Organic Chemistry 2 Study Guide

**Jill Meryl Levy**



## Organic Chemistry 2 Study Guide:

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T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder s tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us [Organic Chemistry, 13e Student Study Guide and Solutions Manual](#)

T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2022-05-03 Organic Chemistry Student Study Guide and Solutions Manual 13th Edition offers the full solutions for select exercises from the text [Study Guide to Accompany Basics for Chemistry](#) Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions

calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students **Study Questions in Elementary Organic Chemistry** Alexander Lowy,1926 **National Library of Medicine Current Catalog** National Library of Medicine (U.S.),1993 First multi year cumulation covers six years 1965 70 *General Register* University of Michigan,1887 Announcements for the following year included in some vols [Calendar of the University of Michigan for ...](#) University of Michigan,1881 **Catalogue of the University of Michigan** University of Michigan,1886 Announcements for the following year included in some vols [Chemistry, Grades 6 - 12](#) Barbara R. Sandall,2009-12-16 Connect students in grades 5 and up with science using Chemistry Physical and Chemical Changes in Matter This 80 page book reinforces scientific techniques It includes teacher pages that provide quick overviews of the lessons and student pages with Knowledge Builders and Inquiry Investigations that can be completed individually or in groups The book also includes tips for lesson preparation materials lists strategies and alternative methods of instruction a glossary an inquiry investigation rubric and a bibliography It allows for differentiated instruction and supports National Science Education Standards and NCTM standards

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