

## Organic chemistry II (alkanoic acids and alkanols)

- Ethylbutanoate
  - $\text{CH}_3\text{CH}_2\text{CH}_2$
  - Esters
- $-\text{CH}-\text{CH}-\text{CH}_2-\text{CH}-\text{CH}_2-\text{CH}-\text{CH}_2-\text{CH}$
  - Polyphenyl ethane
- Plastics may contain chlorine or fluorine compounds apart from hydrogen and carbon when burnt, fluorine and chlorine compounds are released into the air destroying Ozone layer
- $(\text{NH}_4)_2\text{CO}_{(aq)} \quad 2\text{NH}_{3(g)} + \text{CO}_{2(g)} + \text{H}_2\text{O}_{(l)}$
- The first amount of soap precipitates  $\text{Ca}^{2+}_{(aq)}$  and  $\text{Mg}^{2+}_{(aq)}$  ions and soften water. Then additional soap dissolves oil from the fabric.
- $$\begin{array}{ccccccc} \text{CH}_3\text{CH}_2 & \text{O} & & \text{O} & \text{CH}_2\text{CH}_3 & \text{O} & \\ | & & & & | & & \\ -\text{NH}-\text{CH} & - & \text{C} & -\text{NH}-\text{C} & - & \text{C} & - & \text{C}- \end{array}$$

$$\begin{array}{ccc} \text{H} & & \text{H} \\ | & & | \\ \text{b) } 0.00005\text{mol. P} = 0.515 \text{ g of monomer.} \\ - 1.0 \text{ mole of poly mer} = \frac{1 \times 0.515}{0.0005} = 10300 \text{ g} \end{array}$$

$$\begin{aligned} \text{RFM (C}_4\text{H}_9\text{ND}_2)_n &= 48 + 9 + 32 = 103 \\ &= (\text{C}_4\text{H}_9\text{NO}_2) = 10300 \\ 103n &= 10300 \\ \therefore n &= 100 \text{ molecules} \end{aligned}$$

- Agent A – magnesium salt formed is soluble hence doesn't form scum
- (a) Styrene/Phenylethene

(b) Addition polymerization

(c) – can be made into different shapes easily

- are cheaper
  - are not corroded by acids, alkalis or air
  - are stronger and long lasting
  - are water-proof
- Any 1 correct

- Add water to the mixture and shake where ethanol dissolves in water while pentane is immiscible.

\*MAT

- Transfer the mixture in a separating funnel and allow it to settle when pentane floats on top of water-ethanol mixture.

\*MAT

- Turn on the tap to collect water-ethanol mixture while pentane remains in the separating funnel.
- Separate ethanol from water by fractional distillation based on the differences in boiling points.

- Is 100% ethanol/is pure ethanol without water in it
  - 30°C and yeast

✓ 1

- $R = \frac{\Delta x}{\Delta t}$
- $= \frac{43 - 40.5}{}$

# Organic Chemistry Que And Ans

**RD Boyd**



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## **Table of Contents Organic Chemistry Que And Ans**

1. Understanding the eBook Organic Chemistry Que And Ans
  - The Rise of Digital Reading Organic Chemistry Que And Ans
  - Advantages of eBooks Over Traditional Books
2. Identifying Organic Chemistry Que And Ans
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Organic Chemistry Que And Ans
  - User-Friendly Interface
4. Exploring eBook Recommendations from Organic Chemistry Que And Ans
  - Personalized Recommendations
  - Organic Chemistry Que And Ans User Reviews and Ratings
  - Organic Chemistry Que And Ans and Bestseller Lists
5. Accessing Organic Chemistry Que And Ans Free and Paid eBooks
  - Organic Chemistry Que And Ans Public Domain eBooks
  - Organic Chemistry Que And Ans eBook Subscription Services
  - Organic Chemistry Que And Ans Budget-Friendly Options
6. Navigating Organic Chemistry Que And Ans eBook Formats

- ePub, PDF, MOBI, and More
- Organic Chemistry Que And Ans Compatibility with Devices
- Organic Chemistry Que And Ans Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Organic Chemistry Que And Ans
  - Highlighting and Note-Taking Organic Chemistry Que And Ans
  - Interactive Elements Organic Chemistry Que And Ans
- 8. Staying Engaged with Organic Chemistry Que And Ans
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Organic Chemistry Que And Ans
- 9. Balancing eBooks and Physical Books Organic Chemistry Que And Ans
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Organic Chemistry Que And Ans
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Organic Chemistry Que And Ans
  - Setting Reading Goals Organic Chemistry Que And Ans
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Organic Chemistry Que And Ans
  - Fact-Checking eBook Content of Organic Chemistry Que And Ans
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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