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Question type: Multiple choice

- 1) Which of the following compounds contains polar covalent bonds?
- A) CS₂
- B) LiF
- $C) F_2$
- D) CH₃F
- E) None of these choices.

Answer: D

Topic: Polar Covalent Bonds Section Reference 1: 2.2 Difficulty: Easy

2) Which molecule does not have a dipole moment?

$$A)$$
 F

D) F
E) None of these choices.

Answer: D

Topic: Molecular geometry, polarity Section Reference 1: 2.3

Organic Chemistry Solomons 11 Solution

T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder

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