

Practice Final Exam, Chemistry 2220, Organic Chem II

1. Rank the designated protons by chemical shift ( $\delta$ ), highest  $\delta$  first.



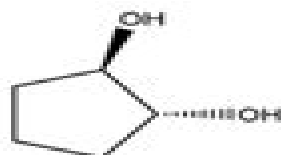
- A. IV > III > II > I    B. IV > II > III > I    C. III > IV > II > I    D. III > IV > I > II

2. What are the products, if any, expected from the following reaction?



- $\text{CH}_3\text{OCH}_3 + \text{NaHS}$      $\text{CH}_3\text{SCH}_3 + \text{NaOH}$      $\text{CH}_3\text{O}-\text{H} + \text{CH}_3\text{S}^-\text{Na}^+$     no reaction  
 A.                                  B.                                  C.                                  D.

3. What is the IUPAC name for the following compound?



- A. cis-1,2-cyclopentanediol  
 B. meso-1,2-cyclopentanediol  
 C. (1R,2R)-1,2-cyclopentanediol  
 D. (1R,2S)-1,2-cyclopentanediol

4. Which of the following explains why cyclohexanol has a  $\text{pK}_a$  of 18 and phenol has a  $\text{pK}_a$  of 10?

- A. Phenolate is a stronger base than the conjugate base of cyclohexanol  
 B. The conjugate base of cyclohexanol is resonance stabilized  
 C. The conjugate base of phenol is resonance stabilized  
 D. Phenol is a weaker acid than cyclohexanol

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### Table of Contents Organic Chemistry 2 Final Exams

1. Understanding the eBook Organic Chemistry 2 Final Exams
  - The Rise of Digital Reading Organic Chemistry 2 Final Exams
  - Advantages of eBooks Over Traditional Books
2. Identifying Organic Chemistry 2 Final Exams
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Organic Chemistry 2 Final Exams
  - User-Friendly Interface
4. Exploring eBook Recommendations from Organic Chemistry 2 Final Exams
  - Personalized Recommendations
  - Organic Chemistry 2 Final Exams User Reviews and Ratings
  - Organic Chemistry 2 Final Exams and Bestseller Lists

5. Accessing Organic Chemistry 2 Final Exams Free and Paid eBooks
  - Organic Chemistry 2 Final Exams Public Domain eBooks
  - Organic Chemistry 2 Final Exams eBook Subscription Services
  - Organic Chemistry 2 Final Exams Budget-Friendly Options
6. Navigating Organic Chemistry 2 Final Exams eBook Formats
  - ePub, PDF, MOBI, and More
  - Organic Chemistry 2 Final Exams Compatibility with Devices
  - Organic Chemistry 2 Final Exams Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Organic Chemistry 2 Final Exams
  - Highlighting and Note-Taking Organic Chemistry 2 Final Exams
  - Interactive Elements Organic Chemistry 2 Final Exams
8. Staying Engaged with Organic Chemistry 2 Final Exams
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Organic Chemistry 2 Final Exams
9. Balancing eBooks and Physical Books Organic Chemistry 2 Final Exams
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Organic Chemistry 2 Final Exams
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Organic Chemistry 2 Final Exams
  - Setting Reading Goals Organic Chemistry 2 Final Exams
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Organic Chemistry 2 Final Exams
  - Fact-Checking eBook Content of Organic Chemistry 2 Final Exams
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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