ALKENES:

General formula = Cn.Hzm

functional group: c=C (savurated)

REACTIONS:

Hydration (#20) -> #-c-c-#

- accidit carryst
- nigh pressurer remp
- Forms account

Halogenarion(802) => 4 - 2-2-4

Hydrogenation(42) => # #

-nriker calabyst

composition =7 complete: (02th20 incomplete: 10tH20

WITCOHOIT:

functional group ON

General Formua = Callenn OH

REACTIONS:

Burning => Product: (02 +420

Add water (420) => (no H or OH)

Add oxygen (02) => forms (almonytic androls

formentation => Ethanol. 4) Time . . filterto ormove
yeast . disni remove water.

CABOXYLIC ACIDS:

functional group = COOH

General formula: ChHzner COOH.

REACTIONS:

Acid+Water => CH3(00H= CH3(00"+H" weak acid-partially lonised "

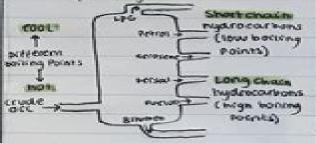
Acid+ (arbonate => (#3(00H+ Nag(03 -> (0g+ H20 + CH3(00NA (ag) due to CO2

ACCOL + ALCOHOL -> CH3COOH + CH3CH2OH -> CH3C COCH3CH2 + HILO ESTEC.

eg ernanou acid+ernanoi > ernyi ernanoate

FRACTIONAL DISTULATION:

vised to apperous musicocar bon fractions.



POLYMERISATION:

Addition Polymerisation -: No side products

Condensation paymerisation . water side module

ROMMON . 40- 6-0-6-04 + H-N-1-1-4 (NN-1-NH)

- [2-0-2-n-d-n-] + MIO

CRACKING

The breaking up the hydrocarbon chains which are too long from fractional distribution into shorter More vietus hydrocarbons.

TWO Types:

O Steam cracking: Vaporised hydrocarbon + Catalyst.

Example:

(2 Hig =7 (2 Hg +66H)4 (2 Hig + 6 Hg

one alkene one alkane.

(6 H12+ (2H6

Organic Compounds Notetaking Guide

Mike Smith

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