

ChemActivity 1

The Nuclear Atom

(What Is an Atom?)

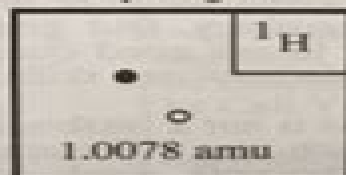
Model: Schematic Diagrams for Various Atoms and Ions.

- electron (-)
- proton (+)
- neutron (no charge)

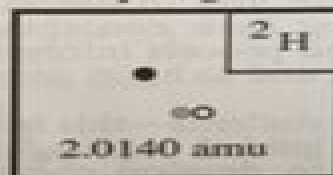
$$1 \text{ amu} = 1.6606 \times 10^{-24} \text{ g}$$

The **nucleus** of an atom contains the protons and the neutrons.

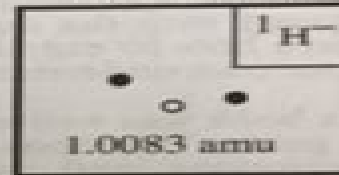
Hydrogen



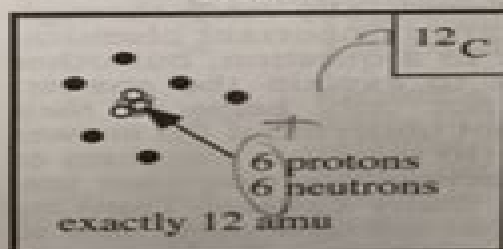
Hydrogen



Hydrogen ion



Carbon



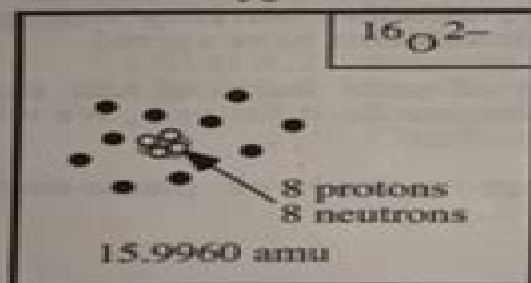
Carbon



Carbon ion



Oxygen ion



Sodium ion



^1H and ^2H are **isotopes** of hydrogen.

^{12}C and ^{13}C are **isotopes** of carbon.

The Shell Model Chemactivity 4

Han-chung Liu



The Shell Model Chemactivity 4:

Chemistry Richard S. Moog, John J. Farrell, 2017-06-26 In the newly updated 7th Edition Chemistry A Guided Inquiry continues to follow the underlying principles developed by years of extensive research on how students learn and draws on testing by those using the POGIL methodology This text follows the principles of inquiry based learning and correspondingly emphasizes underlying chemistry concepts and the reasoning behind them This text provides an approach that follows modern cognitive learning principles by having students learn how to create knowledge based on experimental data and how to test that knowledge

Introductory Chemistry Michael P. Garoutte, Ashley B. Mahoney, 2015-08-10 The ChemActivities found in Introductory Chemistry A Guided Inquiry use the classroom guided inquiry approach and provide an excellent accompaniment to any one semester Introductory text Designed to support Process Oriented Guided Inquiry Learning POGIL these materials provide a variety of ways to promote a student focused active classroom that range from cooperative learning to active student participation in a more traditional setting

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Organic Chemistry, a Guided Inquiry Andrei Straumanis, 2004 *Chemical Abstracts*, 2002 *Prentice Hall Chemistry*, 2000 **Metallurgical Abstracts**, 1960 *Journal of the Institute of Metals* Institute of Metals, 1959 *Decennial Index to Chemical Abstracts*, 1916

Dangerous Properties of Industrial Materials Report, 1996 *Britannica Junior* Harry S. Ashmore, 1961 *Who's who in the world. 1.1971/72(1970)* [Anonymus AC00033125], 1970 **Polymer Science & Technology**, 1969

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The Practitioner's Shell Model George F. Bertsch, 1972 *Dynamics of the Shell Model*, 1985 **The Shell Model of Diatomic Molecules** Han-chung Liu, 1966

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**The Shell Model Chemactivity 4**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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