

Name: Answer Key Class: _____ Date: _____

Periodic Trends Worksheet

1. Classify each of the following elements as a metal, non-metal, or metalloid

- a. Calcium **metal**
- b. Silicon **metalloid**
- c. Chlorine **nonmetal**
- d. Potassium **metal**
- e. Titanium **metal**
- f. Sulfur **nonmetal**
- g. Antimony **metalloid**

2. Determine if the listed pairs of elements have similar or different properties

- a. Potassium & Rubidium **similar**
- b. Calcium & Barium **similar**
- c. Sodium & Chlorine **different**
- d. Helium & Krypton **similar**
- e. Phosphorus & Oxygen **different**

Atomic Radius:

3. What does atomic radius measure? the size of an atom

4. Arrange the following in order of **increasing** atomic radius: (smallest → biggest)

K, Se, Kr, Zn Kr, Se, Zn, K

5. Arrange the following in order of **decreasing** atomic radius: (biggest → smallest)

Pt, At, Ba, W Ba, W, Pt, At

6. Arrange the following in order of **increasing** atomic radius:

Br, At, Cl, F F, Cl, Br, At

7. Arrange the following in order of **decreasing** atomic radius:

Ra, Ca, Be, Ba Ra, Ba, Ca, Be

8. Which is larger? **Ba** atom or Sr atom?

9. Which is smaller? **Al** atom or Te atom?

10. Which is larger? **Zn** atom or Br atom?

11. Which element would you predict to have the largest atomic radius?

Francium

Teaching Transparency Worksheet 21 Periodic Trends Answer

Cindy Blobaum



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