

The Mole

Part 1: Mole and the Molecule

Answer the following questions using the information provided.

- The atomic mass of carbon is 12.011 amu. The mass of 1 mole of carbon is _____ g.
- The atomic mass of oxygen is 15.999 amu. The mass of 1 mole of oxygen is _____ g.
- The mass of 1 mole of H_2O is 18.015 g. The mass of 1 mole of H_2O is _____ g.
- The mass of 1 mole of H_2O is 18.015 g. The mass of 1 mole of H_2O is _____ g.
- The mass of 1 mole of H_2O is 18.015 g. The mass of 1 mole of H_2O is _____ g.

Part 2: Molar Mass

- The mass of 1 mole of H_2O is 18.015 g. The mass of 1 mole of H_2O is _____ g.

10

What is the mass of 1 mole of H_2O ? (The mass of 1 mole of H_2O is 18.015 g.)

18.015

Given that 1 mole = 6.022 $\times 10^{23}$ atoms and 1 mole of H_2O is 18.015 g.

- How many atoms are in 1 mole?

6.022 $\times 10^{23}$

- How many moles are in 18.015 g of H_2O ?

1

Skill Sheet Chapter 1no 1

Clemens Wendtner

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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