Skill Practice 19



1. Write the symbol and charges for the following ions:

Example: calcium v Ca

- A/I pfluospfleide m
- B) magnesises w _____ C) rehidises w

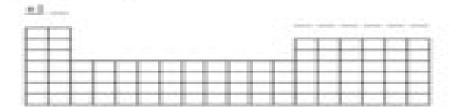
- D) fluoride = E) aluminum = F) sulfide =
- 2. Each of the following formulas is written incorrectly. Please rewrite them correctly. Hyamphr: Ca₂Cl = CaCl₂

 - A) BaS = ______ B) RbN = _____ C) LbC = _____
 - D) Al₂N₁ = _____ F) Mg/Br₂ = ____ F) O₂Al₂ = ____

- 3. Write the formulas for the compound formed by the combination of the following pairs of atoms. Example: nitrogen and magnesium a Ma-N-

A) redison and suffer = _______ B) alaminum and obtoring =

- C) phosphorus and calcium = D) burium and oxygen =
- For each of the formulas you wrote in question 3 above, give the names for the compounds. Example: Mg-N_c w magnesium mbride
- C) D)
- 5. In the blanks above each column, write the charge of the ions formed by atoms in that column. Note: group IA is done for you.



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Kathleen Armour

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