

Program: B.Tech

Course Code & Title: 21CVB101J/Chemistry

Year & Sem: I/II

Date: 29-03-2023

Time: 12.30-1.30 pm

Max. Marks: 30 Marks

Answer ALL the MCQs Part-A (10*1=10 Marks)

1. Which of the following is not a state function
a) ΔG b) ΔE c) W d) H
2. Internal energy does not include
a) Nuclear energy b) Vibrational energy c) Rotational energy d) Energy of gravitational pull
3. Gibbs energy change ΔG is related to equilibrium constant K as:
a) $\Delta G^\circ = -RT \ln K$ b) $\Delta G^\circ = RT \ln K$ c) $\ln K = -RT/\Delta G^\circ$ d) $\ln K = \Delta G^\circ/RT$
4. The conductivity of electrolytic conductors is due to _____
a) Flow of free mobile electrons b) Movement of ions c) Either movement of electrons or ions d) Cannot be said
5. The process of deterioration of a metal due to unwanted chemical or electrochemical interaction of the metal with its environment is called _____
a) Electrolysis b) Electrodialysis c) Corrosion d) Deposition
6. Chemical used in salt bridge is
a) KOH b) KCl c) KNO_3 d) KBr
7. If a compound has 3 chiral carbons What is the number of optically active isomers?
a) 9 b) 3 c) 4 d) 8
8. The potential energy of n-butane is minimum for
a) Skew conformations b) Staggered conformations c) Eclipsed conformations d) Gauche
9. Optical isomerism is a type of _____
a) metamerism b) stereoisomerism c) geometrical isomerism d) tautomerism

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