

ANSWERS - Percent Concentration Practice Worksheet

1. What is the % M/M if 8.0 g copper is added to enough zinc to produce 100 g of an alloy? **8% M/M**

2. Brass is a copper-zinc alloy. If the concentration of zinc is relatively low, the brass has a golden color and is often used for inexpensive jewellery. If a 35.0 g pendant contains 1.7 g of zinc, what is the percentage mass by mass of zinc in brass? **4.9% M/M**

3. What is the % M/M of copper in an alloy when 10 kg of Cu is mixed with 250 kg of Zn? **3.8% M/M**

4. You have 200g of a solution that contains 30g of hydrochloric acid (HCl), what percentage of your solution is made up of hydrochloric acid (M/M)? **15% M/M**

5. If I make a solution by adding water to 75 mL of ethanol until the total volume of the solution is 375 mL, what's the percent by volume of ethanol in the solution? **20% V/V**

6. How would you make 500mL of 30% vinegar in water solution?

30mL vinegar x 500mL sol'n = 150mL

100mL sol'n

So, you would place 150mL of vinegar in a flask and bring to level (500mL) with water.

Skill Practice 39 Concentration Practice Answers

**William J. Kraemer, Steven J.
Fleck, Michael R. Deschenes**



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Harcourt Science: Life science, units A and B, 2002

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