

Section 6-1

Concept Review

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 10 neutrons, and 12 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a octet.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

noble gases are stable while everything
else is missing some

Skills Worksheet Section Simple Ions

Jiyuan Zhang



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Table of Contents Skills Worksheet Section Simple Ions

1. Understanding the eBook Skills Worksheet Section Simple Ions
 - The Rise of Digital Reading Skills Worksheet Section Simple Ions
 - Advantages of eBooks Over Traditional Books
2. Identifying Skills Worksheet Section Simple Ions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Skills Worksheet Section Simple Ions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Skills Worksheet Section Simple Ions
 - Personalized Recommendations
 - Skills Worksheet Section Simple Ions User Reviews and Ratings
 - Skills Worksheet Section Simple Ions and Bestseller Lists

5. Accessing Skills Worksheet Section Simple Ions Free and Paid eBooks
 - Skills Worksheet Section Simple Ions Public Domain eBooks
 - Skills Worksheet Section Simple Ions eBook Subscription Services
 - Skills Worksheet Section Simple Ions Budget-Friendly Options
6. Navigating Skills Worksheet Section Simple Ions eBook Formats
 - ePub, PDF, MOBI, and More
 - Skills Worksheet Section Simple Ions Compatibility with Devices
 - Skills Worksheet Section Simple Ions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Skills Worksheet Section Simple Ions
 - Highlighting and Note-Taking Skills Worksheet Section Simple Ions
 - Interactive Elements Skills Worksheet Section Simple Ions
8. Staying Engaged with Skills Worksheet Section Simple Ions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Skills Worksheet Section Simple Ions
9. Balancing eBooks and Physical Books Skills Worksheet Section Simple Ions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Skills Worksheet Section Simple Ions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Skills Worksheet Section Simple Ions
 - Setting Reading Goals Skills Worksheet Section Simple Ions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Skills Worksheet Section Simple Ions
 - Fact-Checking eBook Content of Skills Worksheet Section Simple Ions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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