

MODEL CHEMISTRY QUESTION PAPER

MODEL QUESTION PAPER

II PUC CHEMISTRY

Time: 3 Hours Max Marks: 90

Instructions: The question paper has four parts A, B, C and D

In part A, each question carries one mark, In part B, each question carries two marks,

In part C, each question carries five marks. In part D, each question carries in D-1, Ten marks & each question in D-2 carries five marks.

PART -A

ANSWER ALL THE FOLLOWING QUESTIONS: 1 x10=10

1. What is an emulsion?
2. State Vanthoffs boyles law.
3. What is a tertiary amine?
4. Name the organic compound formed when 1-propanol vapours are passes through heated copper tube.
5. What is free energy?
6. Give an example for a polysaccharide ?
7. Give an example for a species which shows +I effect.
8. What is a zero order reaction?
9. Give an example for a cationic complex compound.
10. What is an Ellingham diagram?

PART-B

ANSWER ANY TEN OF THE FOLLOWING 2 x10 =20

11. Calculate the Magnetic moment of Fe^{3+} ion.
12. How is methane converted into ethane?
13. Calculate the number of particles per unit cell in FCC.
14. Using electron Gas theory, Explain lustre of metals.
15. Explain geometrical isomerism with an example.
16. Equilibrium constant for a reversible reaction carried due at 25°C is 4.2×10^{-3} . Calculate free energy change for the reaction.

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