

Syllabus for Chemistry 103: General Chemistry I Spring 2012

Class meeting times: MWF 11:00 AM-12:00 PM

Class meeting location: Park Science, Room 278

Required materials:

Text: "Chemistry & Chemical Reactivity" (8th Edition) by John C. Kotz, Paul M. Treichel, and John Townsend, Brooks/Cole, 2012.

Supplemental: OWL (Online Web Learning) access. OWL will be used for completion of all problem set assignments. See <http://www.cengage.com/owl> and the course Moodle site for more details.

Course description: This course is the first semester of a comprehensive survey of modern descriptive, inorganic, and physical chemistry. Topics include: atomic theory, stoichiometry, thermochemistry, chemical periodicity, concepts in chemical bonding, the shapes of molecules, carbon-based molecules, the properties of gases, and intermolecular forces.

Course objectives: We will cover Chapters 1-12 in the textbook this semester, with the possible exception of Chapter 10. Specific learning objectives for each chapter are included at the beginning of each textbook chapter and will be noted in lecture.

Instructor: Prof. Michael Lawlor

Office: Park Science 171

Phone: 610-526-5136

E-mail: mdlawlor@brynmawr.edu

Office hours: Prof. Lawlor's office hours are Monday and Wednesday, 2:30-4:00 PM, Thursday, 10:00-11:30 AM, or by appointment (which is best arranged by e-mail).

Email: Every effort will be made to respond to urgent matters in a timely fashion, but students may not always receive an email reply until the next day.

Course policies

Academic integrity: Integrity is a crucial part of the academic experience. Academic integrity is a core institutional value at Bryn Mawr College, and I take it very seriously in this course. I expect you to follow the Bryn Mawr College Honor Code and work independently on exams and quizzes.

Attendance: It is very important to attend lecture. Your grade will reflect your class attendance as lecture and discussion material and problems will form the basis of exam and homework questions.

Moodle: You are responsible for the information posted on the Moodle course site. Course documents such as the syllabus, homework assignments, and answer keys will be posted on Moodle. If an exam must be rescheduled or a class cancelled, these Course Announcements will also be posted on Moodle.

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The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 2 is a concise and accessible textbook that covers the critical information a student needs to understand the basic mathematics used in chemistry courses. The book provides easy to understand step by step instructions for solving general chemistry problems. The book begins with chapters dedicated to solutions, kinetics, and liquids, solids, and phase changes. In subsequent chapters, the text covers important topics like equilibrium, concentrations, strong and weak acids and bases, the Common Ion Effect, and reaction mechanisms. It also covers the equilibrium between a solid and its respective ions in a solution, as well as the second law of Thermodynamics. The text also addresses Gibbs Free Energy, equilibrium constants, and electrolysis calculations. Each chapter contains sample problems and practice problems to help further understanding of how math and chemistry go hand in hand. The Answer Key is an excellent resource for any undergraduate course that deals with the basic concepts of general chemistry.

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Chemistry Answer Key Units 6-10 (RES) Responsive Educations Solutions Staff, 2012-08-01 **Student's key : answers for all questions and even-numbered problems in "Chemistry, matter, and the universe"** Richard Earl Dickerson, Irving Geis, 1976 **Marthe Villalonga**, 1978 Answer Key to Experimental Chemistry and Workbook, 1969

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The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Rachel Turoscy,2017-12-31 The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 2 is a concise and accessible textbook that covers the critical information a student needs to understand the basic mathematics used in chemistry courses The book provides easy to understand step by step instructions for solving general chemistry problems The book begins with chapters dedicated to solutions kinetics and liquids solids and phase changes In subsequent chapters the text covers important topics like equilibrium concentrations strong and weak acids and bases the Common Ion Effect and reaction mechanisms It also covers the equilibrium between a solid and its respective ions in a solution as well as the second law of Thermodynamics The text also addresses Gibbs Free Energy equilibrium constants and electrolysis calculations Each chapter contains sample problems and practice problems to help further understanding of how math and chemistry go hand in hand The Answer Key is an excellent resource for any undergraduate course that deals with the basic concepts of general chemistry Rachel Turoscy earned her B S M S and Ph D in chemistry from Lehigh University She is an associate professor of chemistry at Middlesex County College Dr Turoscy has authored numerous articles book chapters and textbooks addressing various topics in the world of chemistry Other Cognella titles by Rachel Turoscy The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 1 First Edition **Assessments Answer Key for Use with Chemistry** Christopher D. Coyle,Kelly Driskell,2021 The assessments answer key includes copies of the student tests and quizzes with overprint answers for easy grading Answers to

printed tests help teachers to assess their students knowledge and understanding of key concepts The answer key includes full worked out solutions to problems Answer keys include page references for the teacher Publisher

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