

Chem 105 Practice Exam Chapters 6, 5, 7 and 8 OpenStax

1) Of the following, which atom has the smallest atomic radius? 14p _____
 A) Mg B) Sr C) Te D) S

2) Which is the longest bond? 12p _____
 A) N-N
 B) N=N
 C) N≡N
 D) All three bond lengths should be about the same.

3) Of the following elements, which has the lowest electronegativity? 18p _____
 A) Ca B) Mg C) Cl D) Br

4) Which molecule contains the most polar bond? 19p _____
 A) CF₄ B) CO₂ C) CH₄ D) CCl₄

5) In the most acceptable Lewis structure for carbonyl fluoride, COF₂ the central atom is 20p _____
 A) O, which is doubly-bonded to C.
 B) C, which is singly-bonded to O.
 C) O, which is singly-bonded to C.
 D) C, which is doubly-bonded to O.

Draw Lewis structure here. (2 pts)

6) Which element can accommodate more than eight electrons in its valence shell? 21p _____
 A) F B) He C) C D) O

7) How many resonance structures are present in the Lewis structure of CO₃²⁻? 22p _____
 A) five B) two C) four D) three

Draw Lewis structure here for one resonance form. (2 pts)

8) Assign formal charges to all atoms in the following resonance form for HNO₂. 23p _____



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**Thirumalaisamy P.
Velavancorresponding**



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