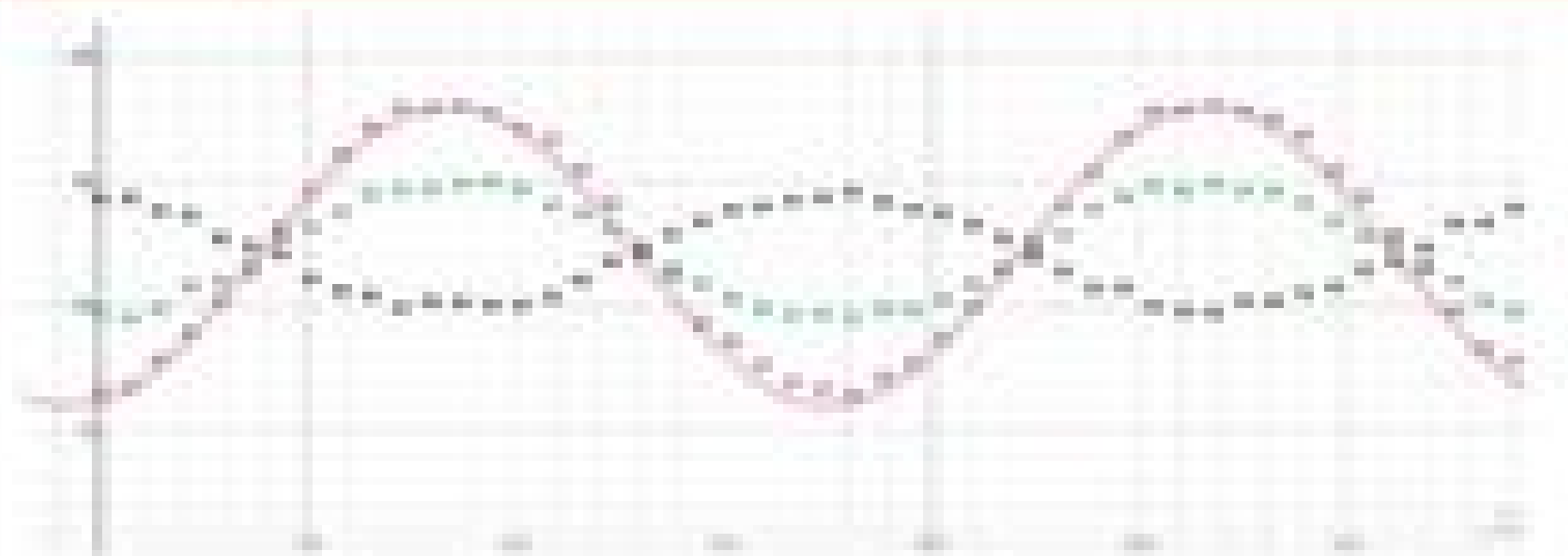


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web problem 1 which of the following are not acids ch<sub>3</sub>cooh co<sub>2</sub> hno<sub>2</sub> hcooh cc<sub>4</sub> h<sub>3</sub>ccooh c<sub>2</sub>hno<sub>2</sub> hccooh ccl<sub>4</sub> shazia naz numerade educator 02 23

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web 2 at what  $pH$  or range of  $pH$  values is a solution considered strongly acidic mildly acidic neutral mildly basic and strongly basic provide a brief explanation

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web 11 why is  $HCl$  a strong acid and  $HClO$  a weak acid 12 why are  $HCl$  and  $HClO$  4 both strong acids 13 for each of the reactions below classify the reactants as an acid or a

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web answer i lemon and orange ii tomatoes and guava iii sour milk curd iv tamarind question 18 a student detected the  $pH$  of four unknown solution a b c and d as